



Application of extraction chromatographic resins in the separation of actinides

S. Happel – TrisKem International

SANDA – workshop on actinide target preparation and characterization – the need for radioanalytical chemistry

Geel, 06/12/2023



TrisKem International



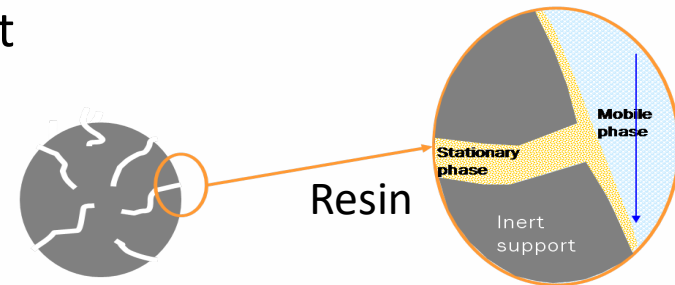
- Based in Rennes (France)
- Independent company since 02/07
 - Formerly part of Eichrom Europe
 - ISO 9001 since 2007
- Main product: extraction chromatographic resins
- Staff : 20
- R&D, QC and TechSupport group:
 - 4 RadChem PhD, 3 Technicians
- R&D: Development of new resins, techniques and applications
- Products used in several domains



Organic extractant impregnated onto inert support

➤ « Supported Solvent Extraction » / « Solvent Impregnated Resins »

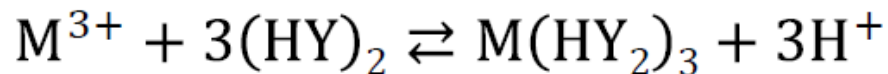
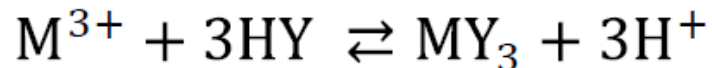
- Distribution between two non-miscible phases
- Stationary phase impregnated onto inert support
 - Choice of inert support depending on application
 - Radiolysis stability, plastic scintillators,...
- High density of functional groups
- Fast kinetics/small volumes => rapid separations
- High variety of selectivities:
 - Pure extractants, synergetic mixtures, solid extractants in diluents
 - **Aim: selectivity for analyte(s), no selectivity for matrix/ impurities**
- Combining several cartridges can improve/facilitate separation
- Bleeding might need to be addressed (e.g. Am on TRU)



Types of Extractants

Acidic

e.g. DIPEX (AC Resin)



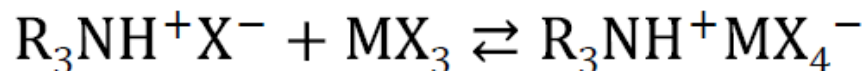
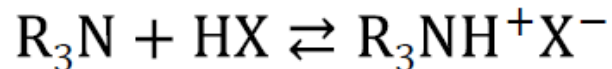
Neutral

e.g. CMPO/TBP (TRU Resin), DPPP (UTEVA Resin), TK221, 'TK200'

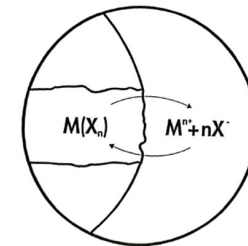


Basic

e.g. Aliquat 336 (TEVA Resin)



Metal Anion Complex Formation

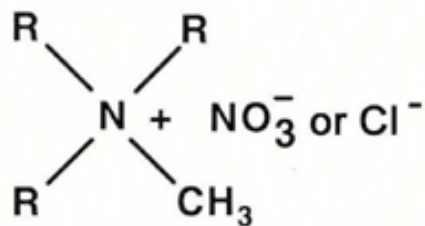


Metal + Anion \rightleftharpoons Complex

Complex + Organic \rightleftharpoons Extracted

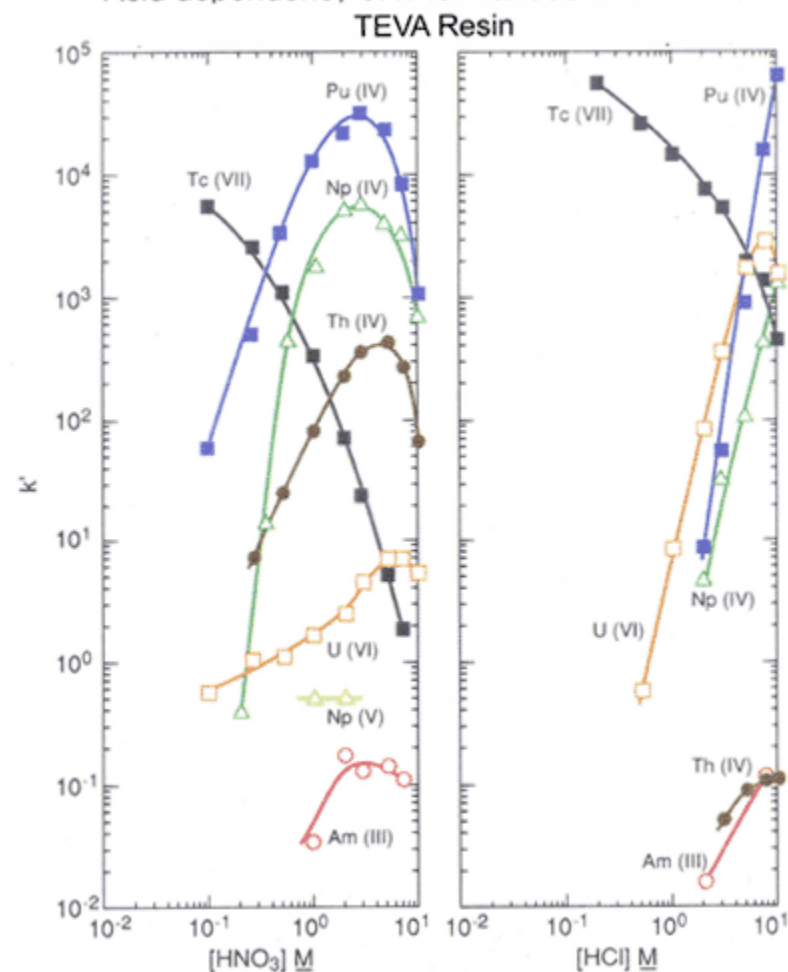
The classics - TEVA Resin

Trialkyl, methylammonium
nitrate (or chloride)



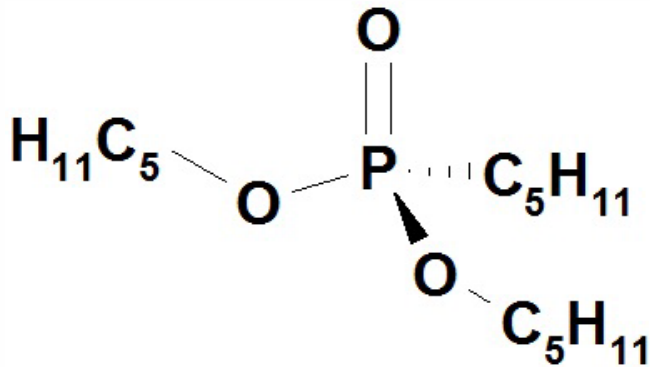
- **Extractant:** Aliquat 336[®]
- TEVA: **TE**t**ra**V**al**ent **A**ctinides
- Pu(IV), Th(IV), Np(IV), Tc(VII),...
- Np/Pu separation
- In HNO₃ no selectivity for U, Am,...
- No selectivity for Fe in HNO₃

Acid dependency of k' for various ions at 23°C.



Horwitz, et al. (HP195)

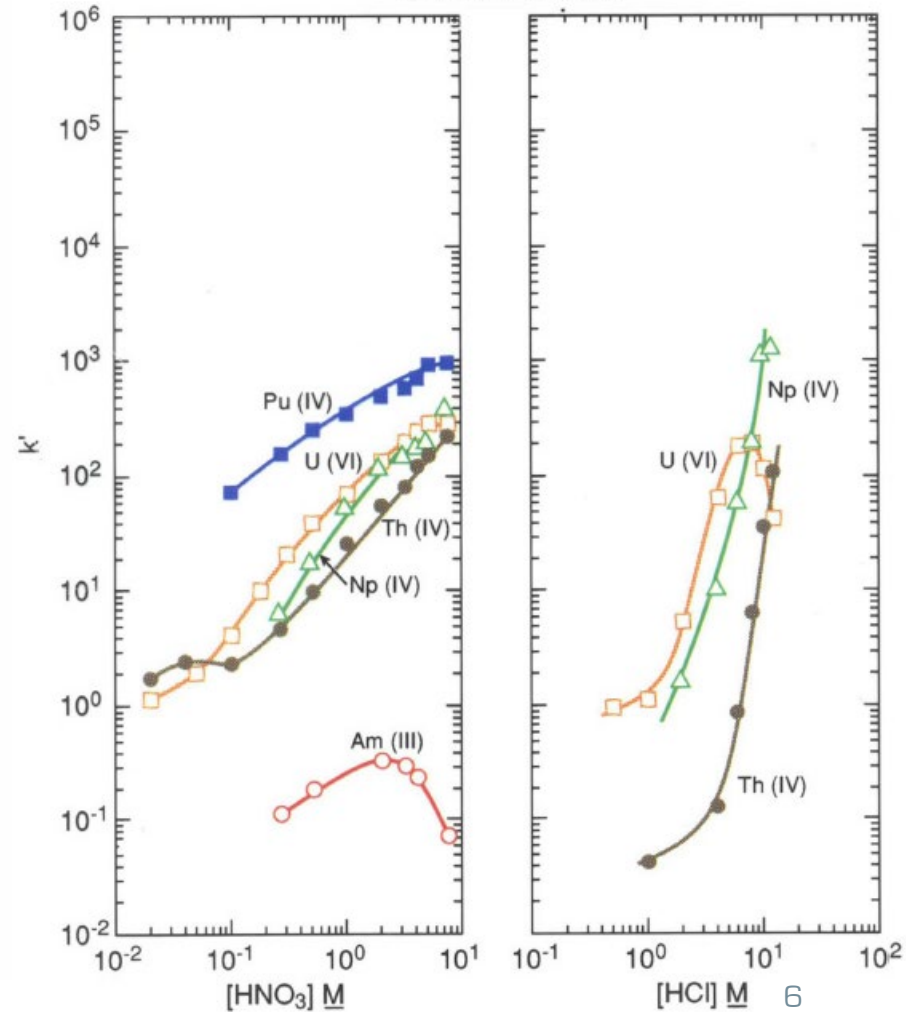
The classics - UTEVA Resin



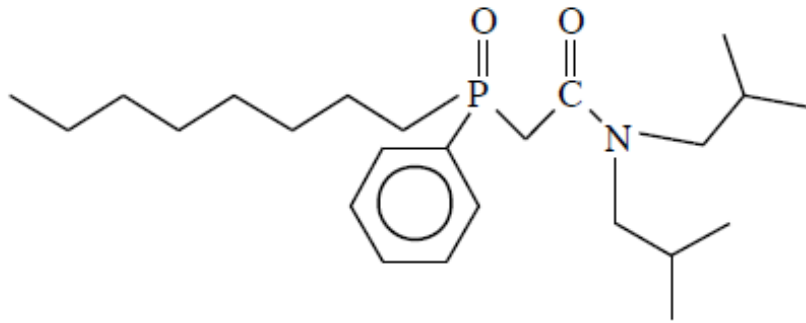
Di-Pentyl-Pentyl-Phosphonate (DPPP)

- **Extractant:** DPPP (in literature also called DAAP)
- For **U**ranium and **TE**ttra**V**alent **A**ctinides
- Good selectivity but U and Th retention not very high
- No interference by Fe

Acid dependency of k' for various ions at 23-25°C.
UTEVA Resin



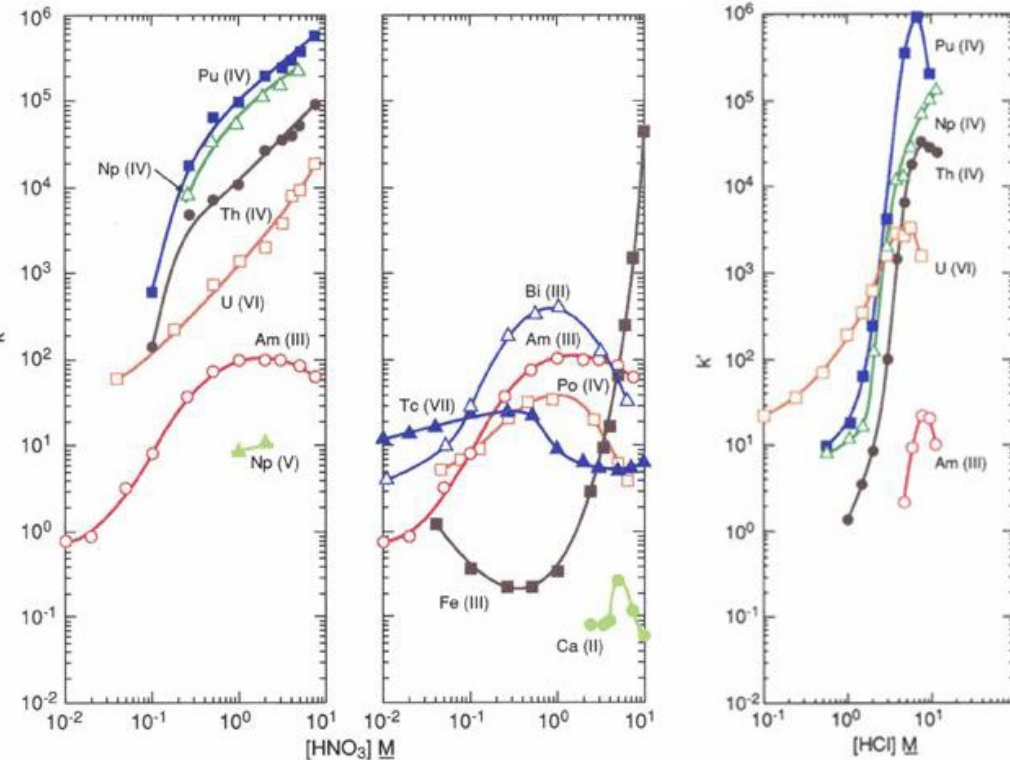
The classics - TRU Resin



octyl(phenyl)-N,N-diisobutylcarbamoylmethylphosphine oxide

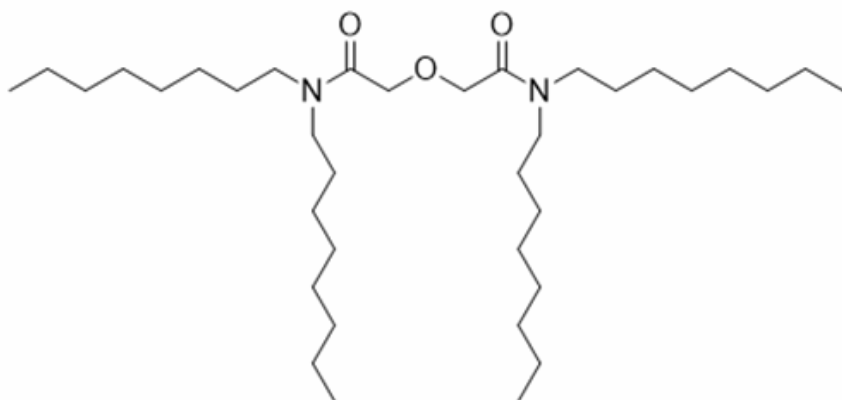
- **Extractant:** CMPO in TBP
- **TRansUranium** elements
- Am/Cm, U, Pu,...
- Am retention interfered by Fe(III)

Acid dependency of k' for various ions at 23-25°C.
TRU Resin

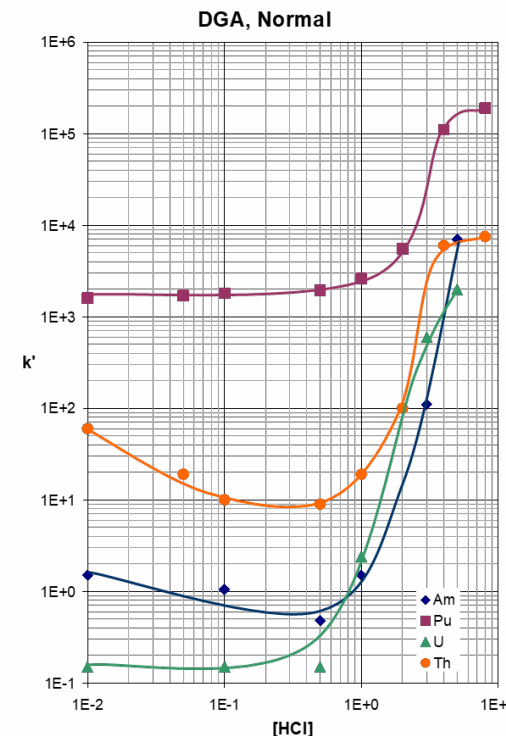
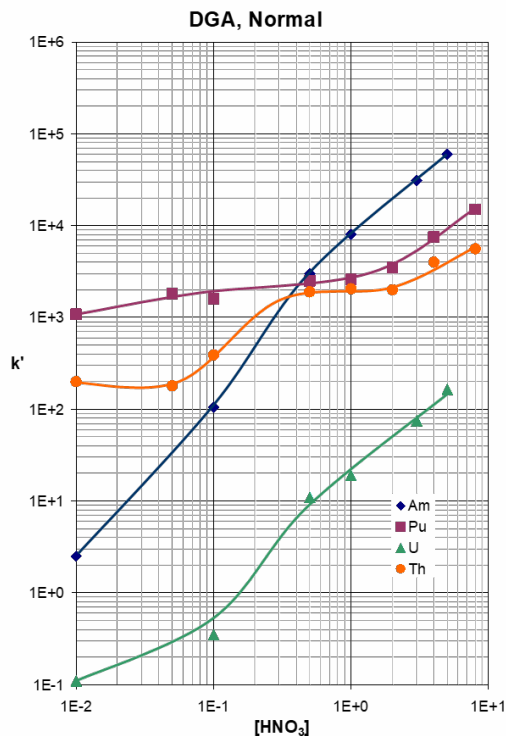


Horwitz, et al. (HP193)

The classics - DGA Resin

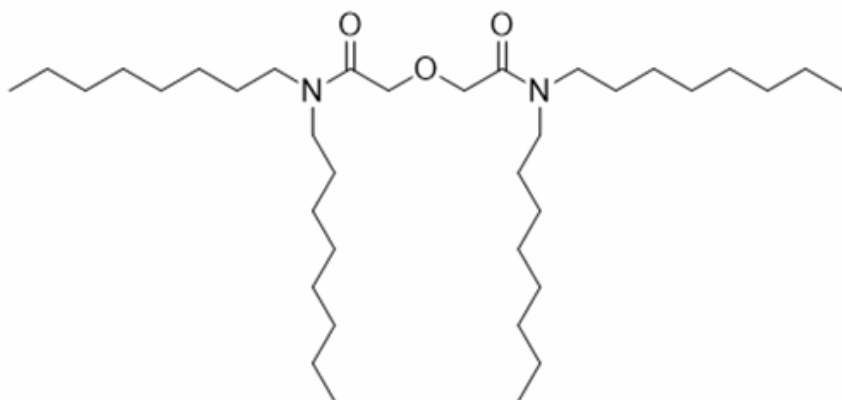


N,N,N',N'-tetra-n-octyldiglycolamide



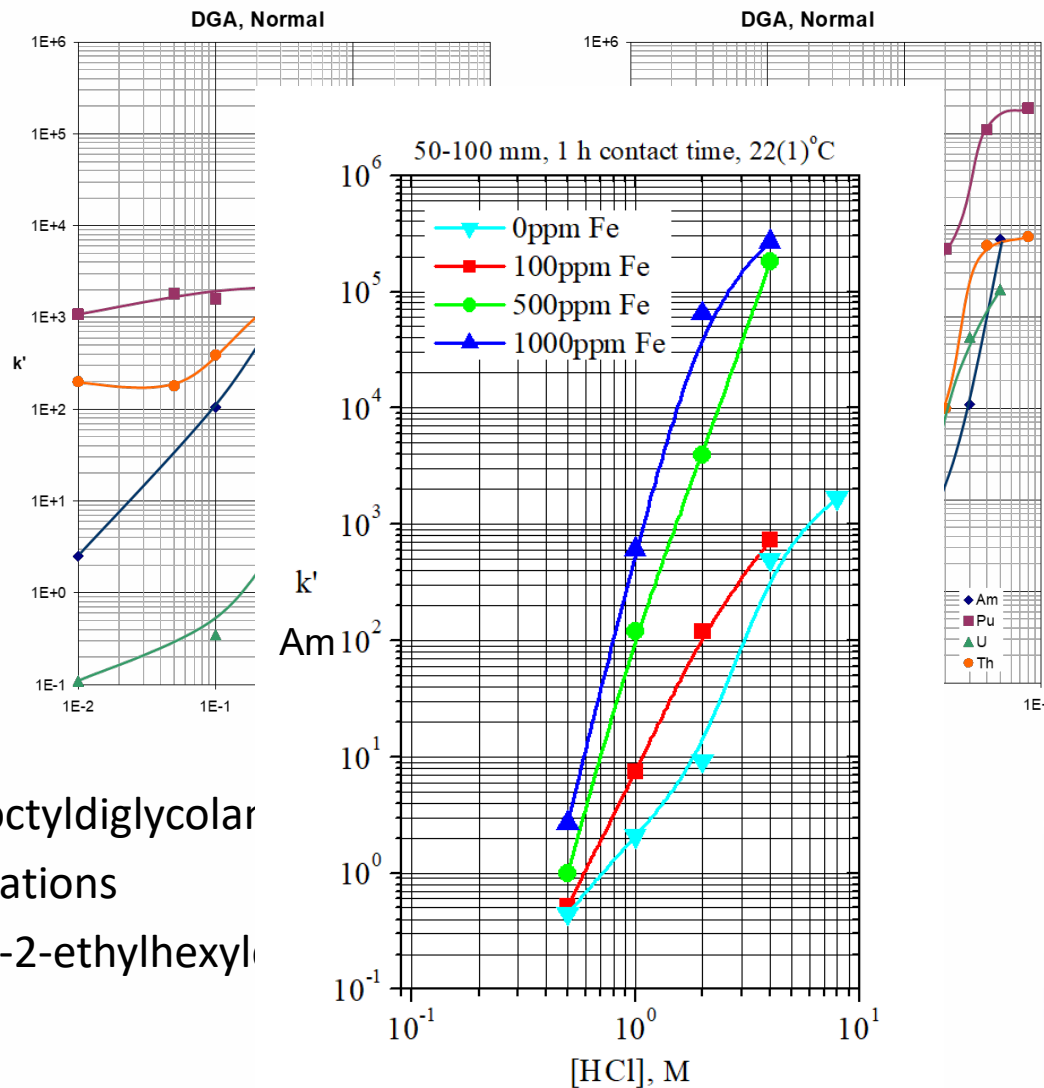
- Extractant:
 - **DGA, Normal** (N,N,N',N'-tetra-n-octyldiglycolamide) = DN
 - Generally used in analytical applications
 - DGA, Branched (N,N,N',N'-tetrakis-2-ethylhexyldiglycolamide) = DB
- Actinides, especially Am
- Robust against Fe interference, often used for soil samples

The classics - DGA Resin



N,N,N',N'-tetra-n-octyl-diglycolamide

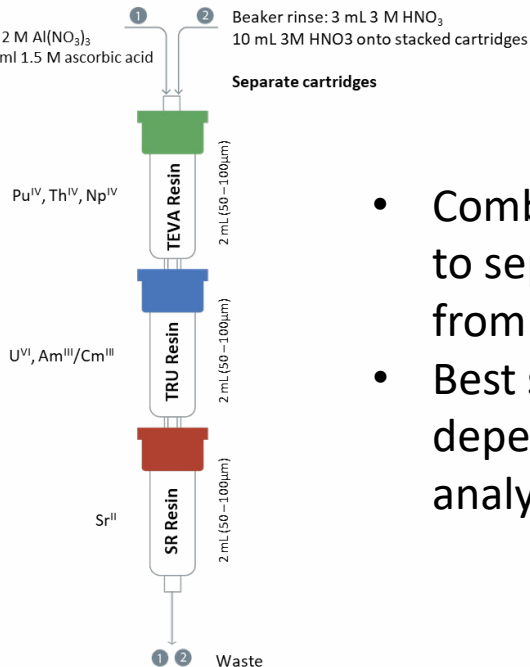
- Extractant:
 - **DGA, Normal** (N,N,N',N'-tetra-n-octyldiglycolar
 - Generally used in analytical applications
 - DGA, Branched (N,N,N',N'-tetrakis-2-ethylhexyl)
- Actinides, especially Am
- Robust against Fe interference, often used



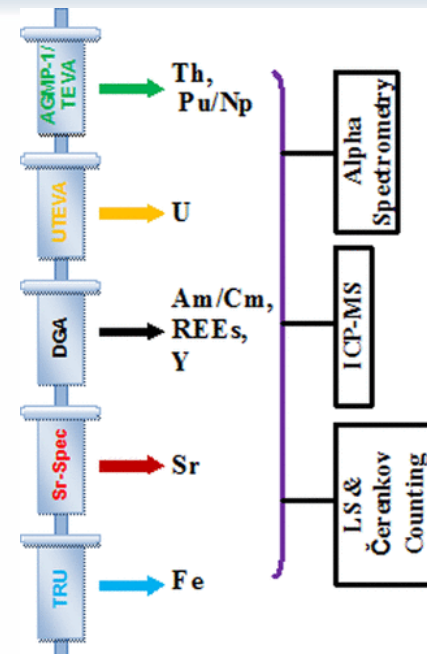
Stacked cartridges

Acid digestion

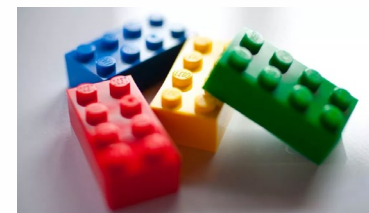
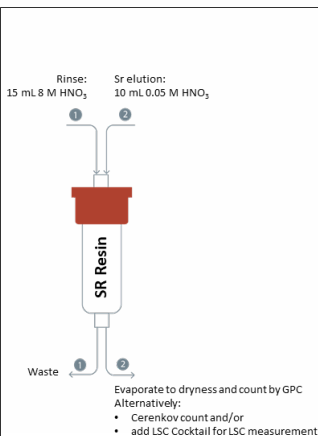
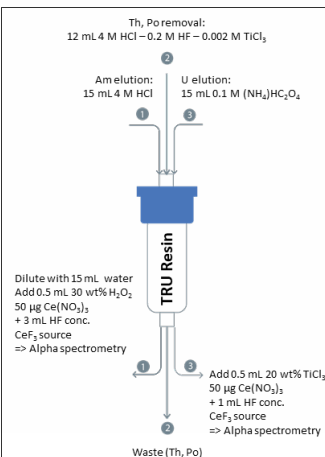
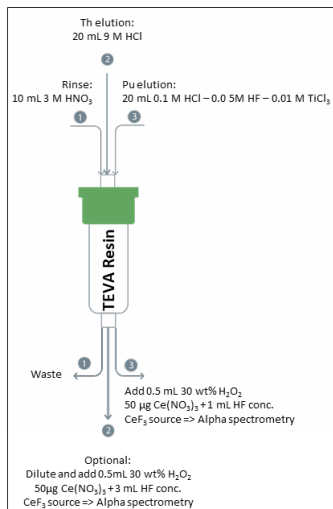
- 1) Redissolve in 6 ml 6 M HNO₃ and 6 ml 2 M Al(NO₃)₃
- 2) Add 0.5 ml 1.5 M sulfamic acid + 1.25 ml 1.5 M ascorbic acid
- 3) Add 1.25 ml 3.5 M sodium nitrate



- Combination of several resins to separate several analytes from one sample
- Best suitable combination depends on matrix (Fe,...) analytes (Am, Np/Pu)



•Xiongxin Dai, Sheila Kramer-Tremblay, Anal. Chem. 2014, 86, 11, 5441–5447, Publication Date: May 6, 2014, <https://doi.org/10.1021/ac500572g>



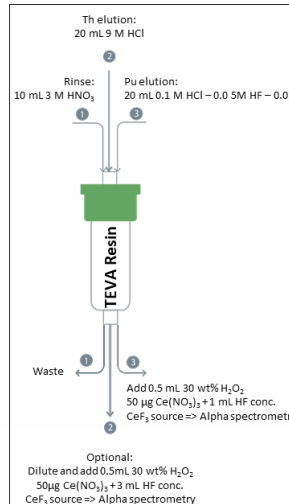
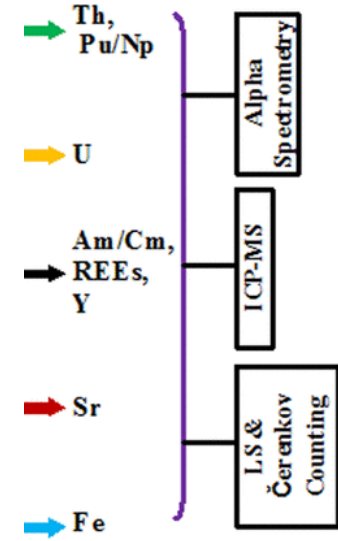
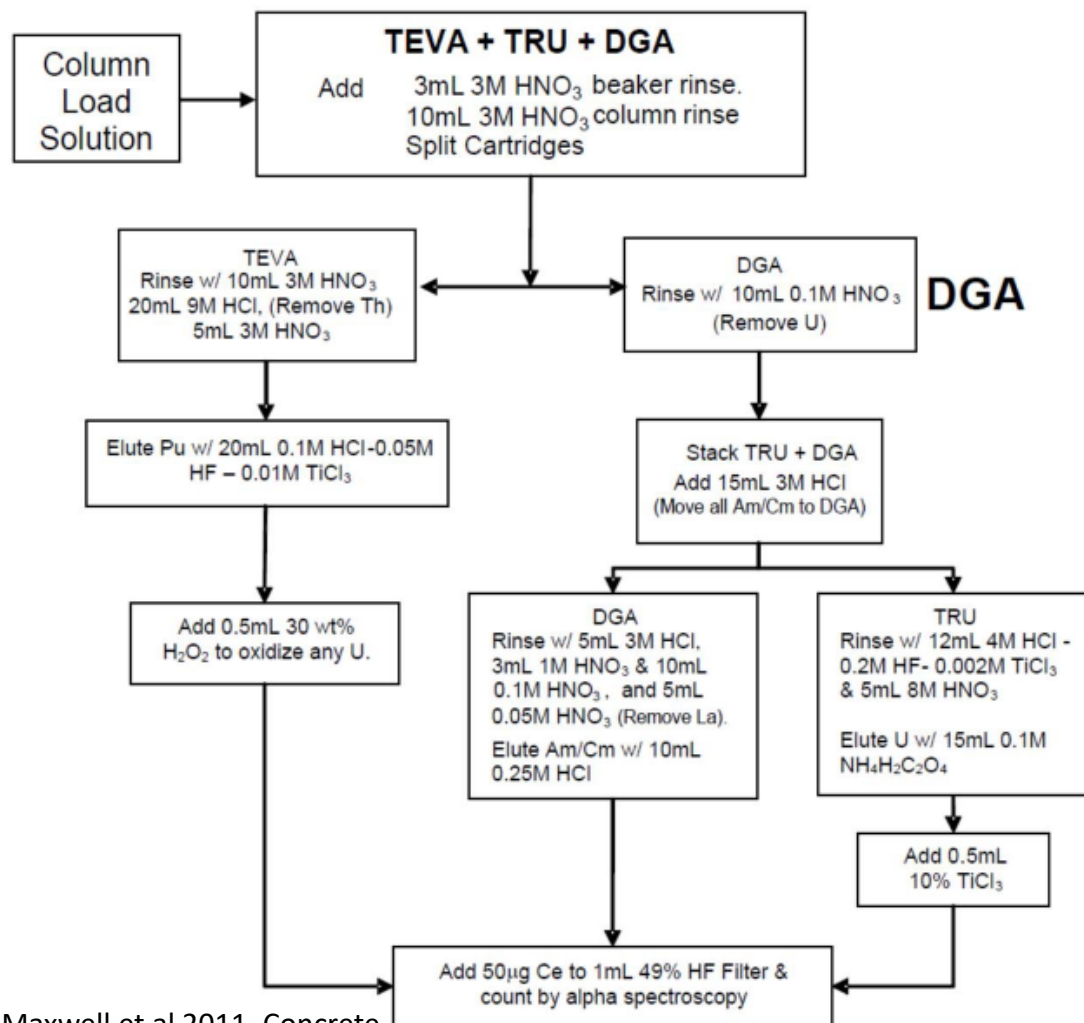
Taken from: Rösch, Frank. Volume 1+2 [Set Rösch: Nuclear- And Radiochemistry, Vol 1+2, De Gruyter, 2023. Referring to Maxwell et al. 2017

Stacked cartridges

Acid digestion

- 1) Redissolve in 6 ml 6 M HNO_3 and 6 ml 2 M $\text{Al}(\text{NO}_3)_3$
- 2) Add 0.5 ml 1.5 M sulfamic acid + 1.25 ml 1.5 M cerium(IV) nitrate
- 3) Add 1.25 ml 3.5 M sodium nitrate

- 1 Beaker rinse: 3 mL 3 M HNO_3
- 2 10 mL 3M HNO_3 onto stacked cartridges



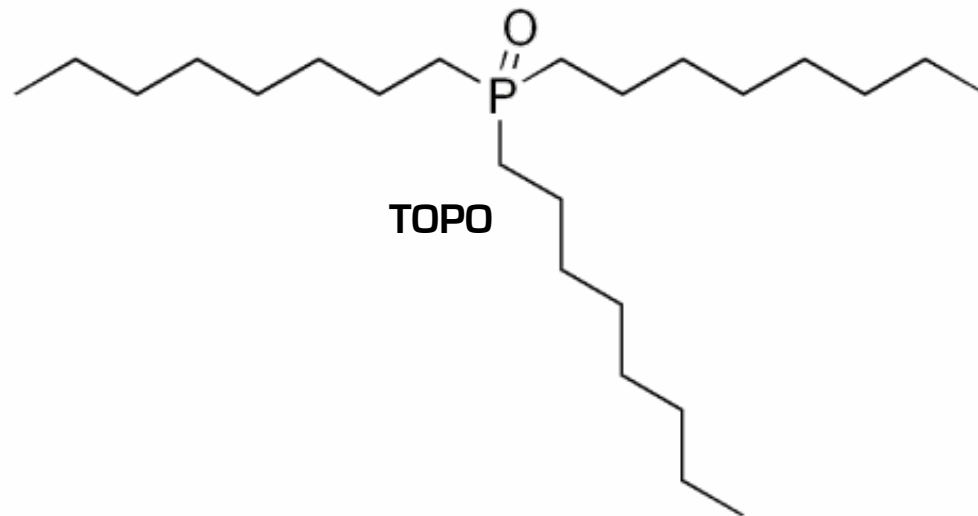
Tremblay, Anal. Chem. 2014, 86, 11, 1166-1172
May 6, 2014, 572g



Maxwell et al 2011, Concrete and brick samples

112, DC (1997), 2023. Referring to Maxwell et al. 2017

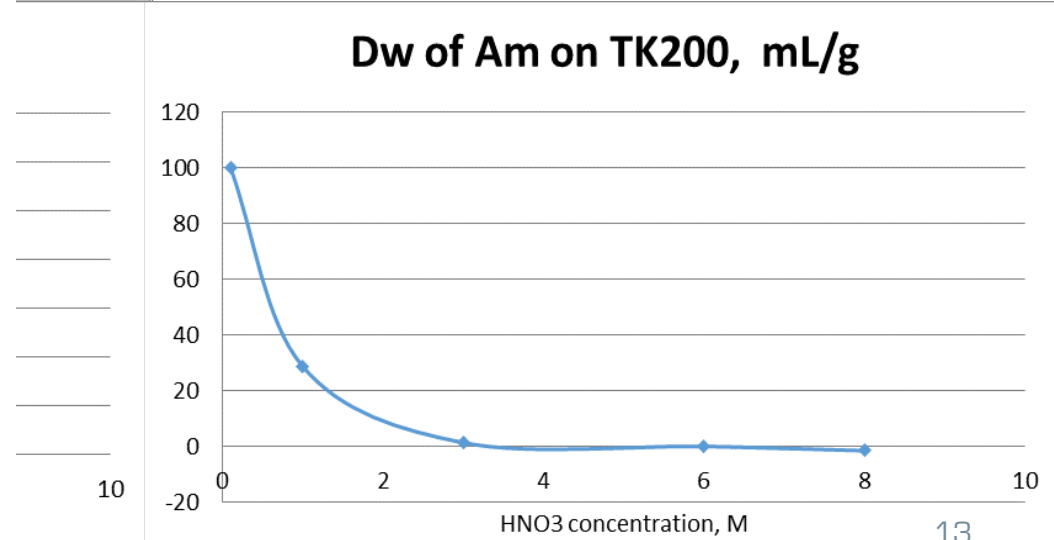
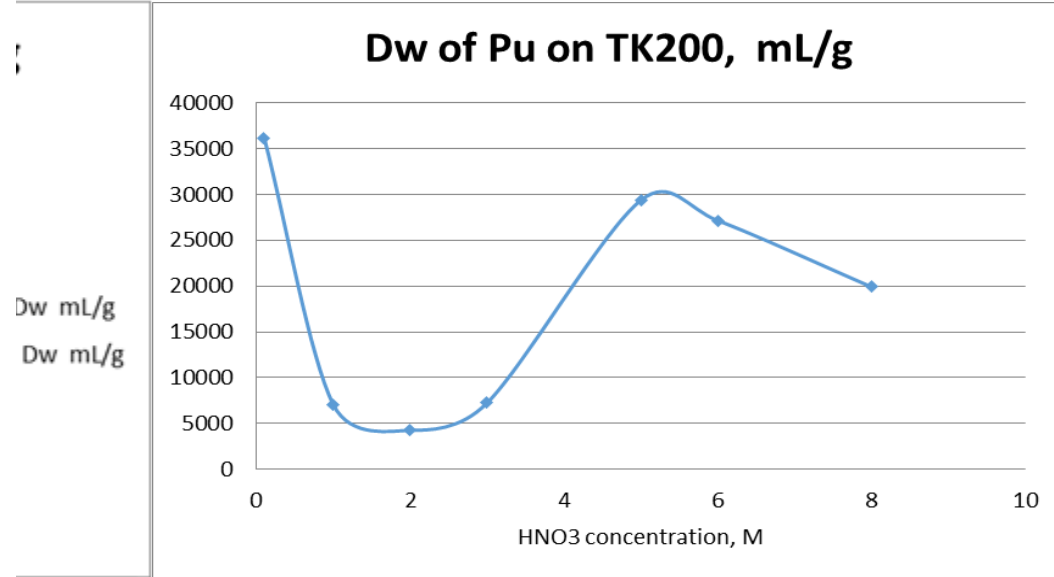
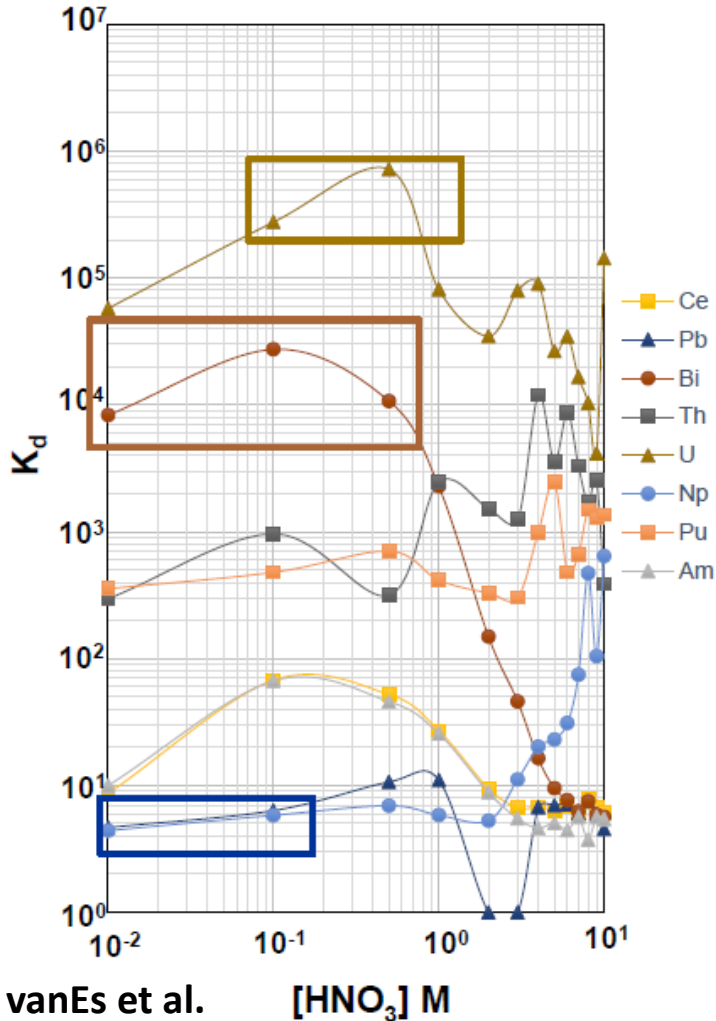
'New' resins - TK200 Resin



- Based on TOPO extractant
- High retention of actinides
 - U, Th higher than on UTEVA
 - In HNO_3 some retention of trivalents
- Extracts actinides even at pH 1 - 2 (nitric acid)
 - Preconcentration and purification of selected actinides on same column
 - 'In the field'?

Actinides on TK200 – HNO₃

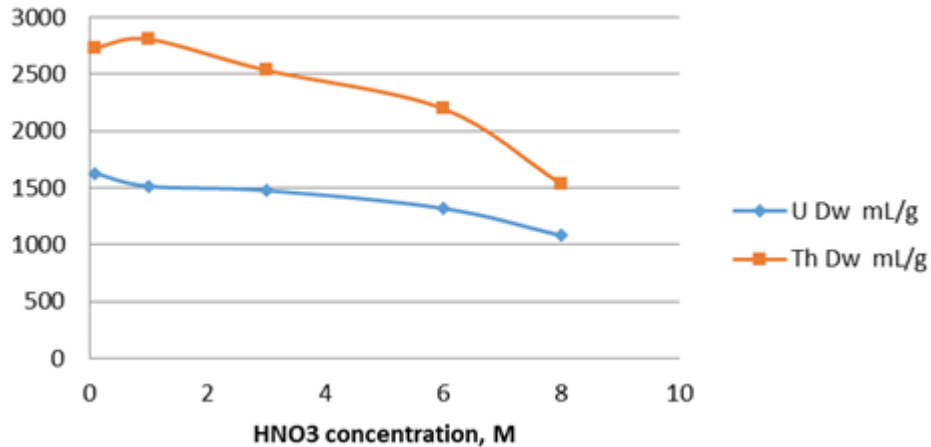
(all data N. Vajda et al)



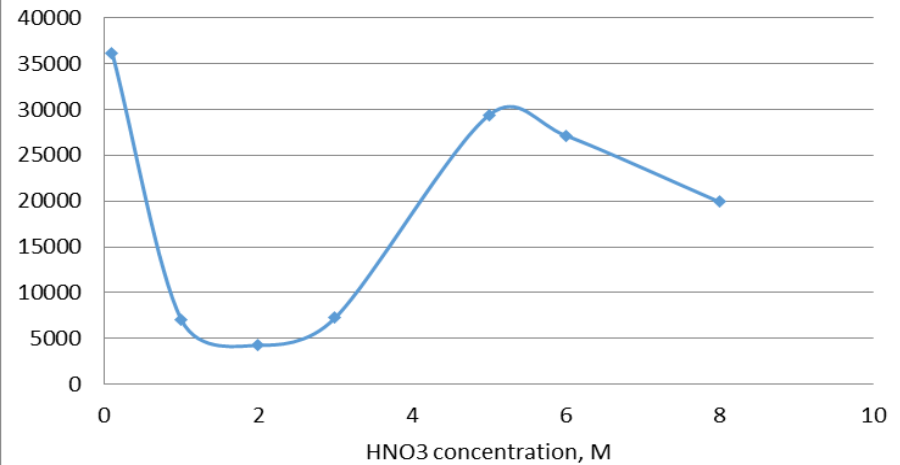
Actinides on TK200 – HNO₃

(all data N. Vajda et al)

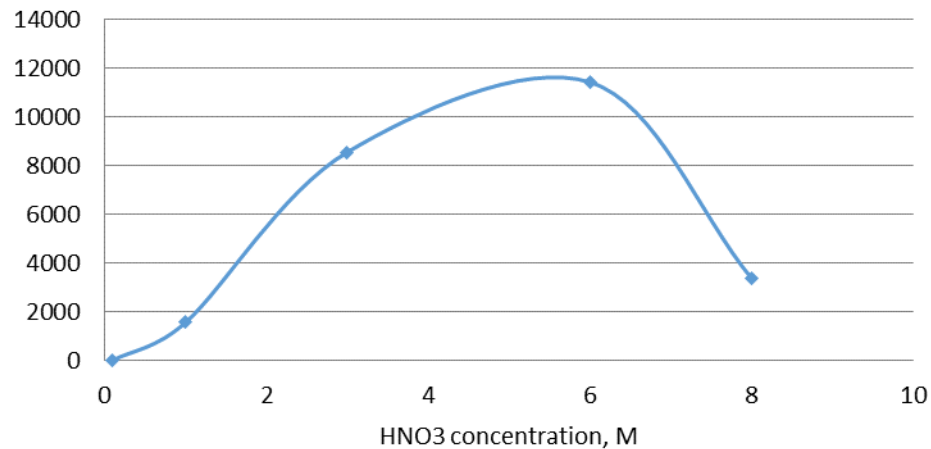
Dw of U(VI) and Th on TK200, mL/g



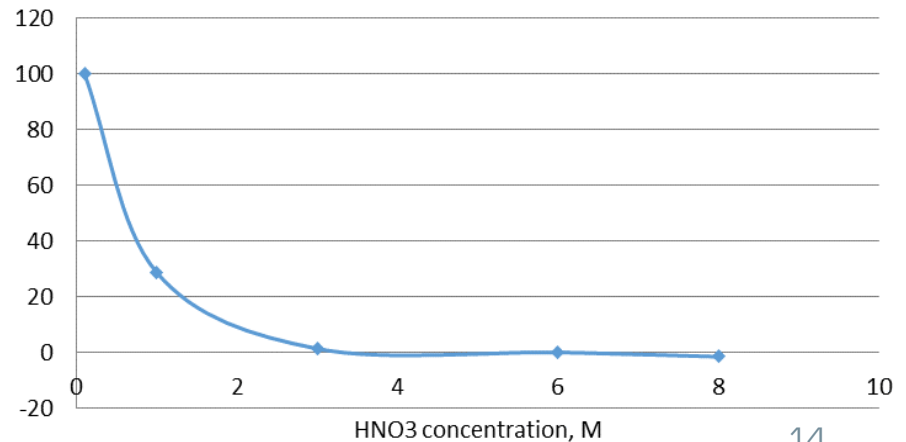
Dw of Pu on TK200, mL/g



Dw of Np(IV) on TK200, mL/g

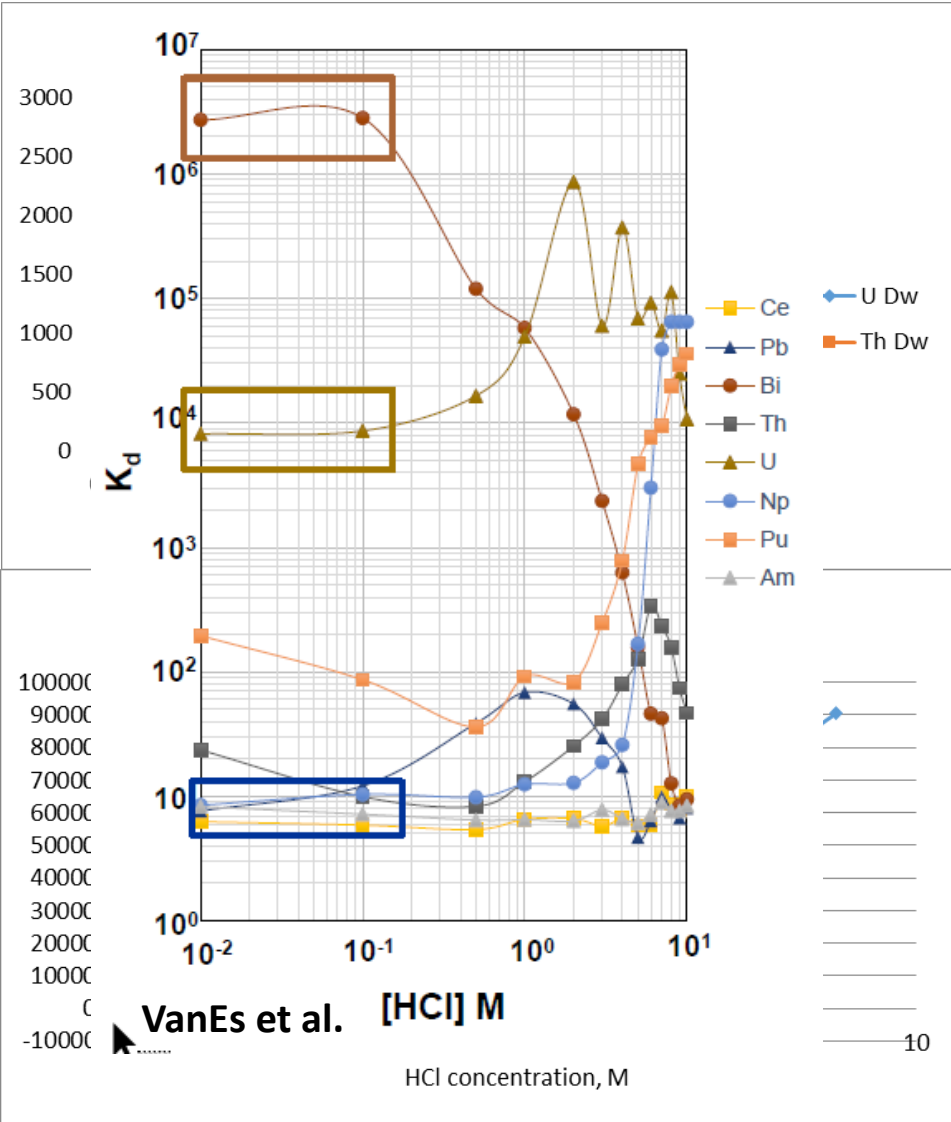


Dw of Am on TK200, mL/g

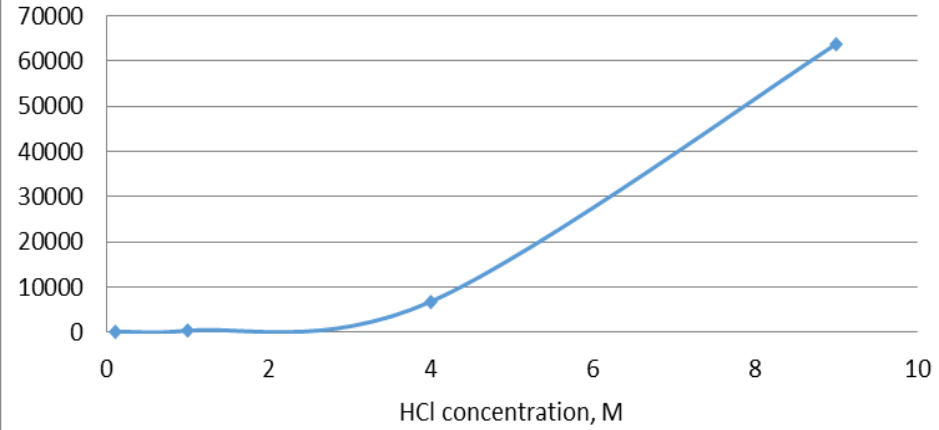


Actinides on TK200 – HCl

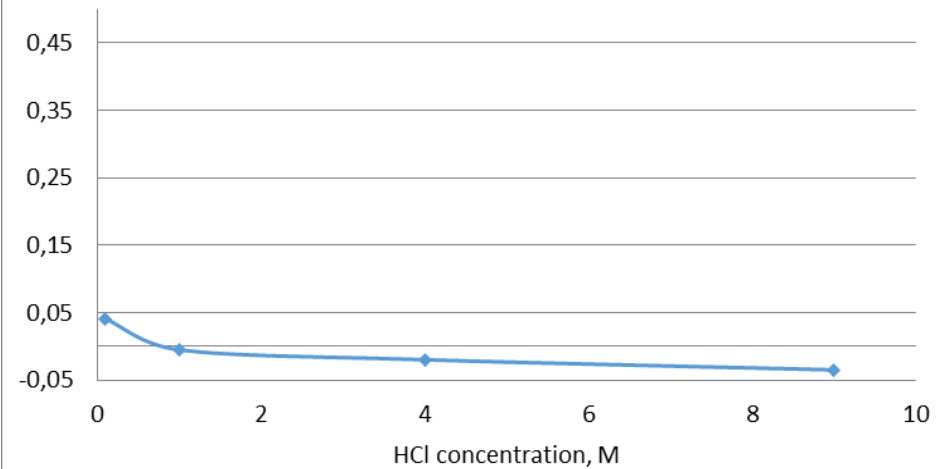
(all data N. Vajda et al)



Dw of Pu(IV) on TK200, mL/g



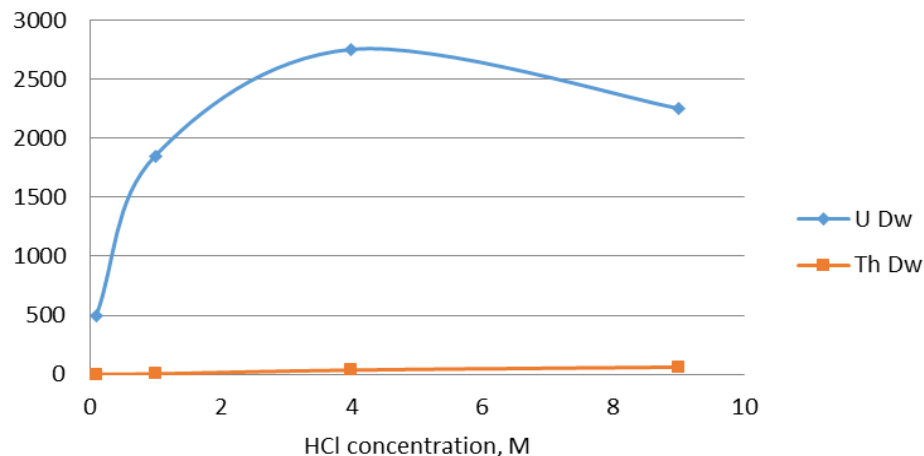
Dw of Am on TK200, mL/g



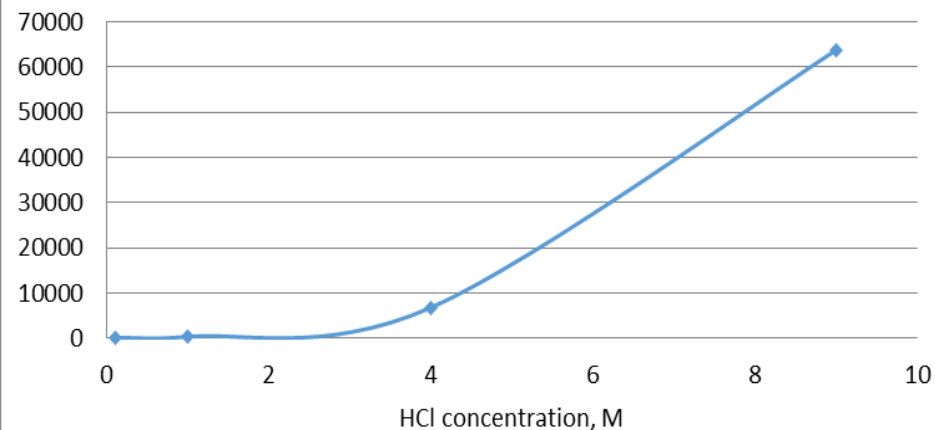
Actinides on TK200 – HCl

(all data N. Vajda et al)

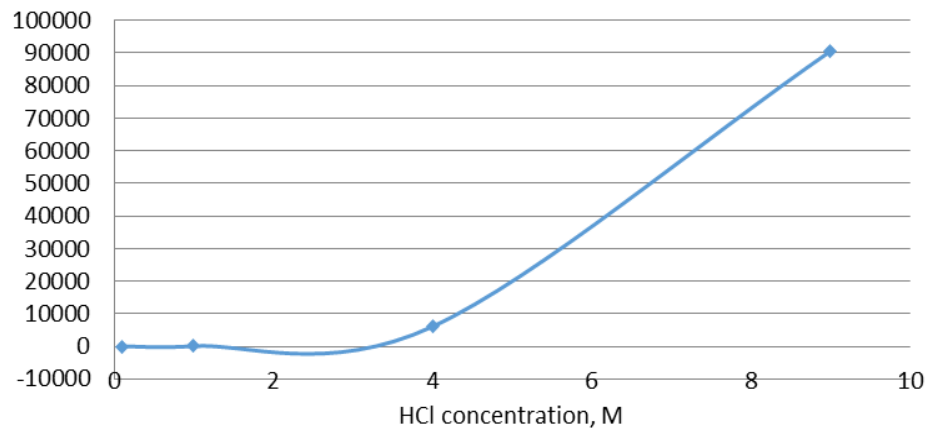
Dw of U(VI) and Th on TK200, mL/g



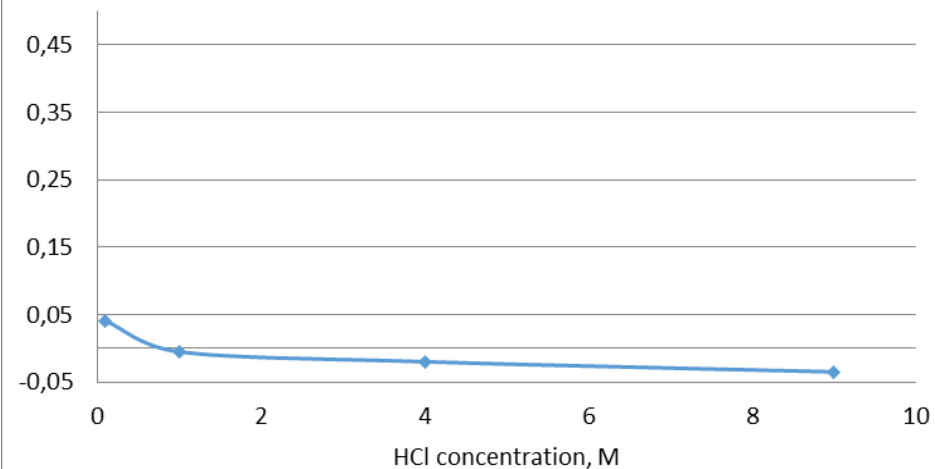
Dw of Pu(IV) on TK200, mL/g



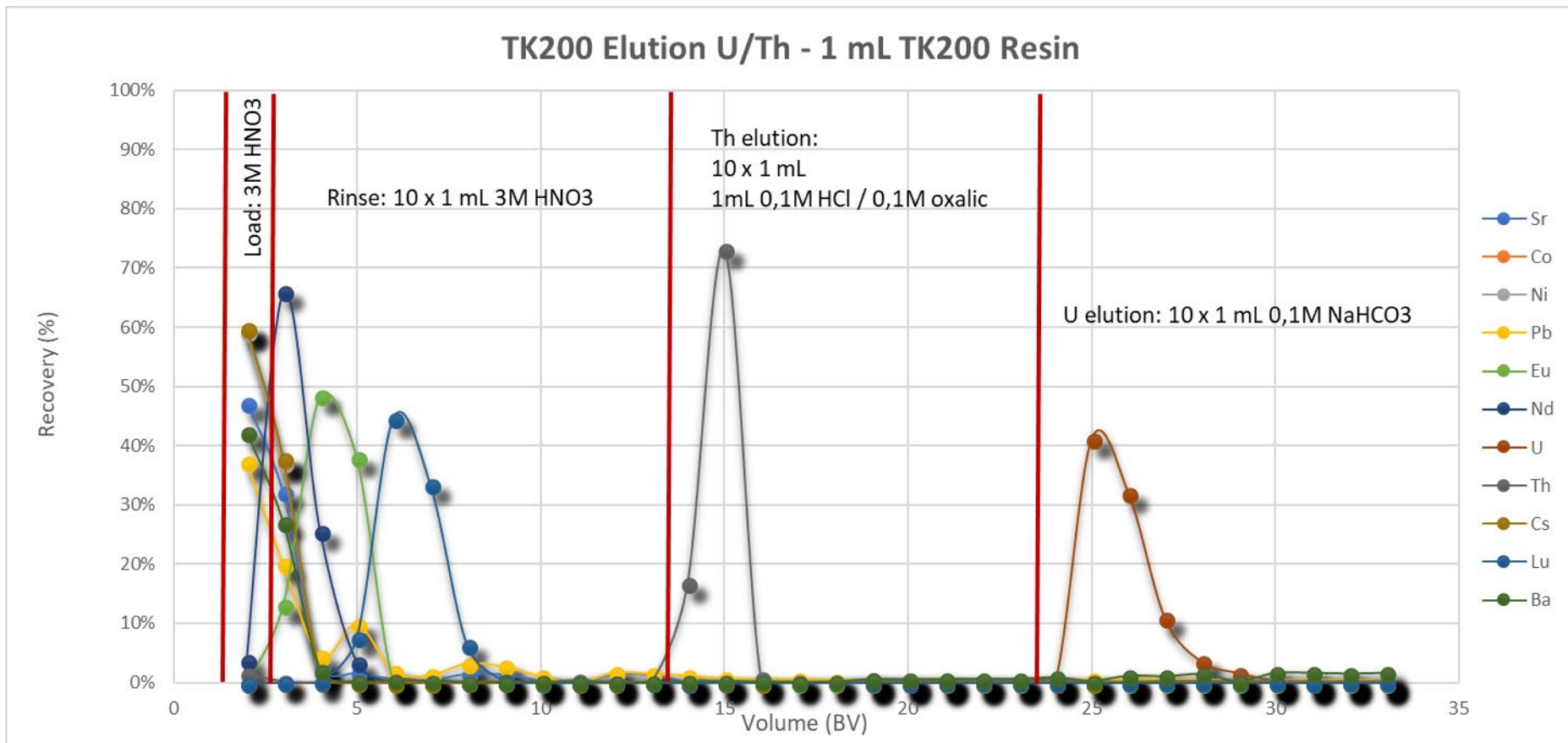
Dw of Np(IV) on TK200, mL/g



Dw of Am on TK200, mL/g

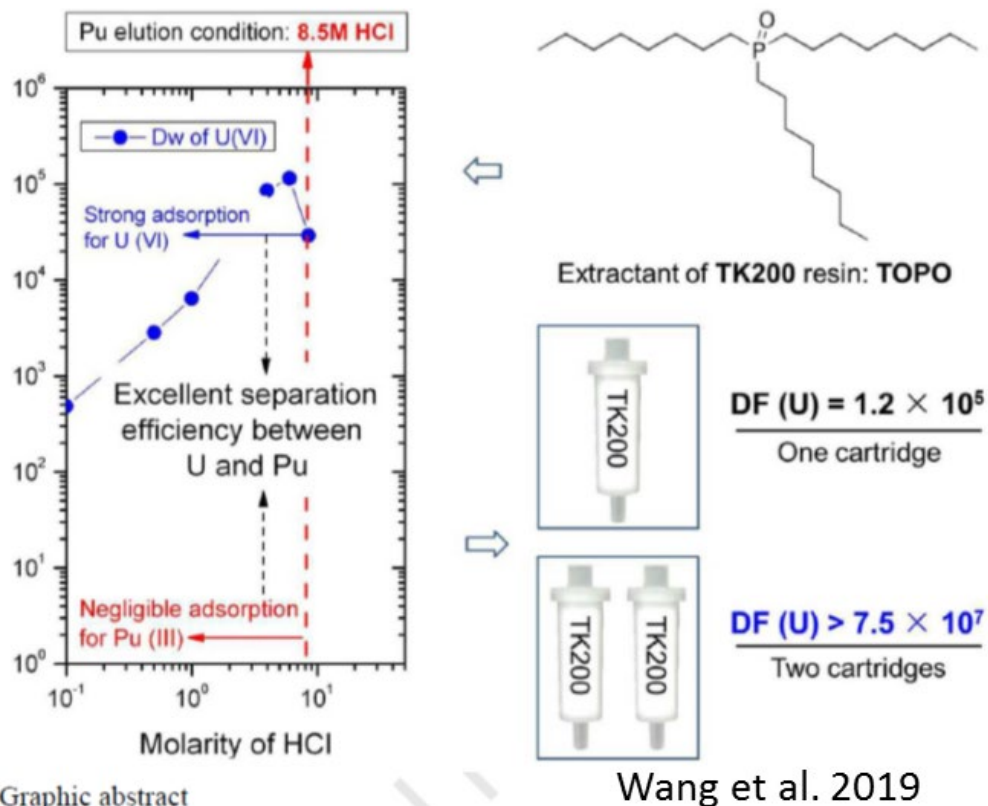


U/Th separation on TK200



- Load: 3M HNO₃ or ≥ 1 L pH2 (HNO₃)
- Very clean U/Th separation
- Oxalate instead of carbonate

TK200 Resin - U/Pu separation - I



Graphic abstract

- Confirms very high U retention
- Pu isotope ratios by ICP-MS
- U removal via 2 x TK200 => Df > 10⁷
- Pu elution as Pu(III) in 8,5M HCl/0,1M NH₄I => U remains fixed on resin in these conditions

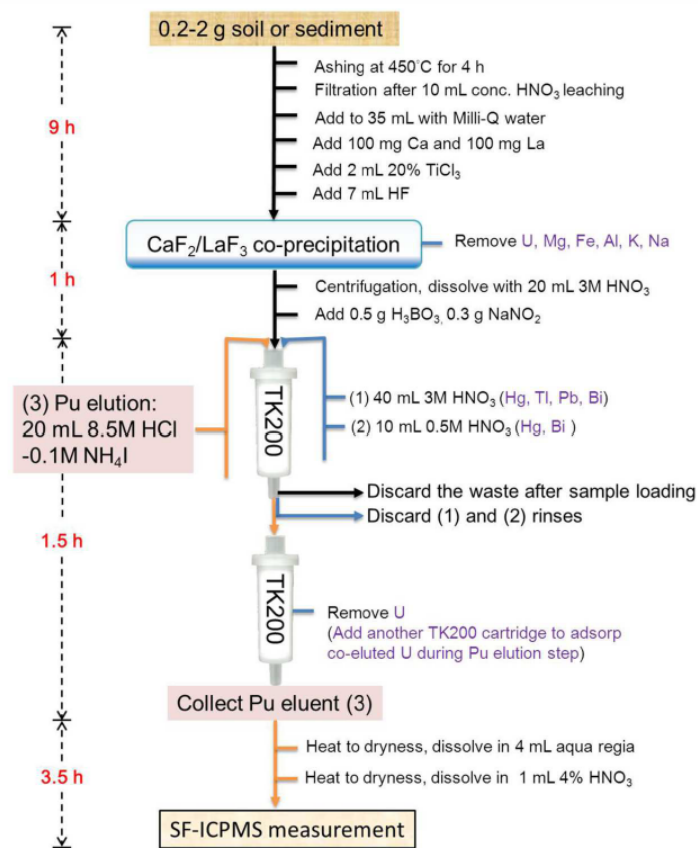
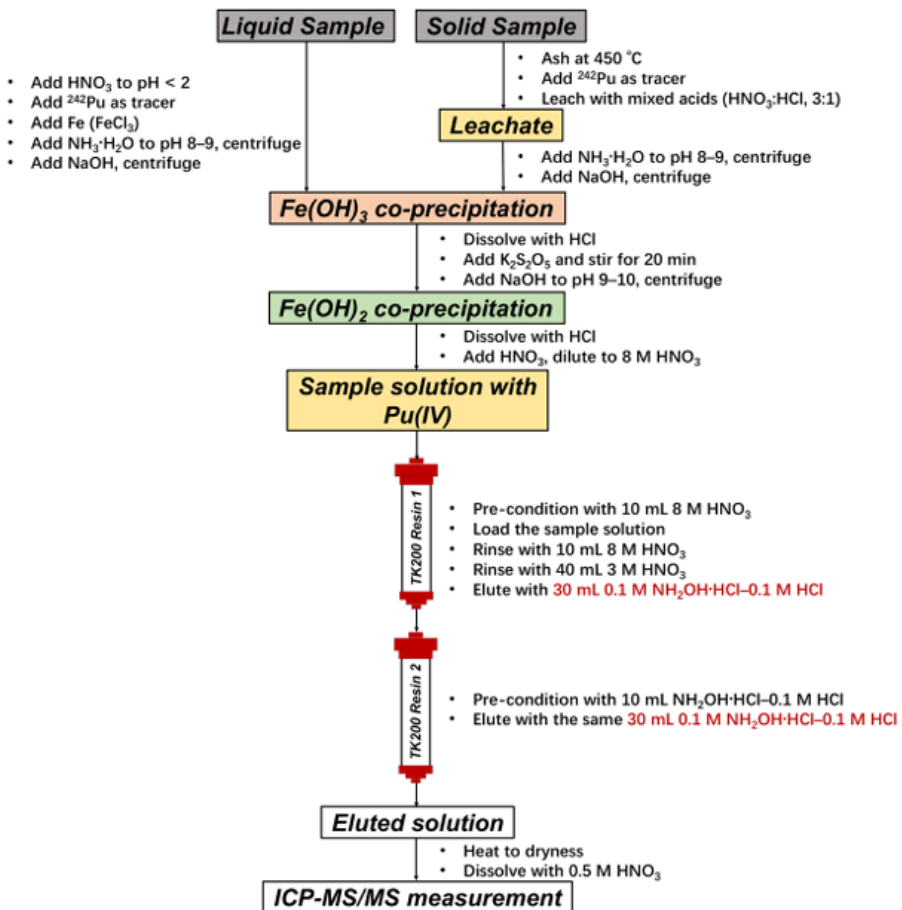


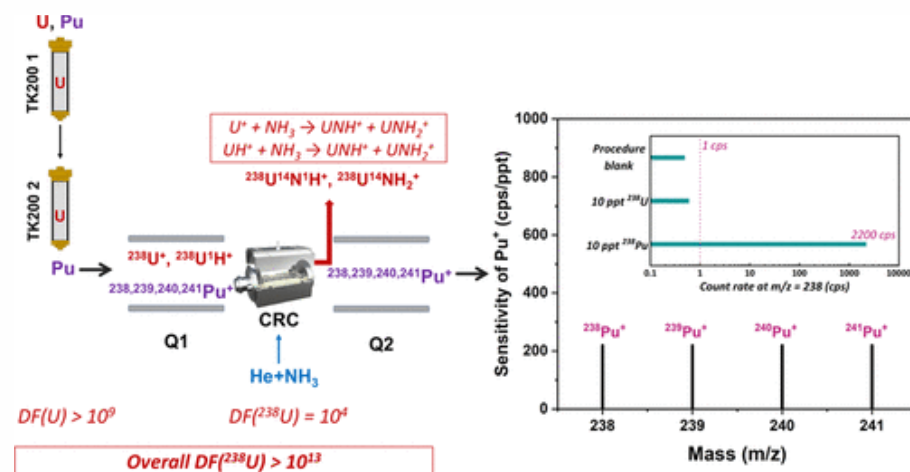
Figure 1

Zhongtang Wang, Zhaoya Huang, Yun Xie, et al. Method for determination of Pu isotopes in soil and sediment samples by inductively coupled plasma mass spectrometry after simple chemical separation using TK200 resin, *Analytica Chimica Acta*, 1090, 2019, 151-158, www.sciencedirect.com/science/article/pii/S000326701931030X

TK200 Resin - U/Pu separation - II



- Recent publication by Huang et al.
- Better U removal: $D_f(U) > 10^9$
- Additional U removal via He+NH₃
- Overall $D_f(U) > 10^{13}$
- Pu isotopes incl. Pu-238 via ICP-MS/MS

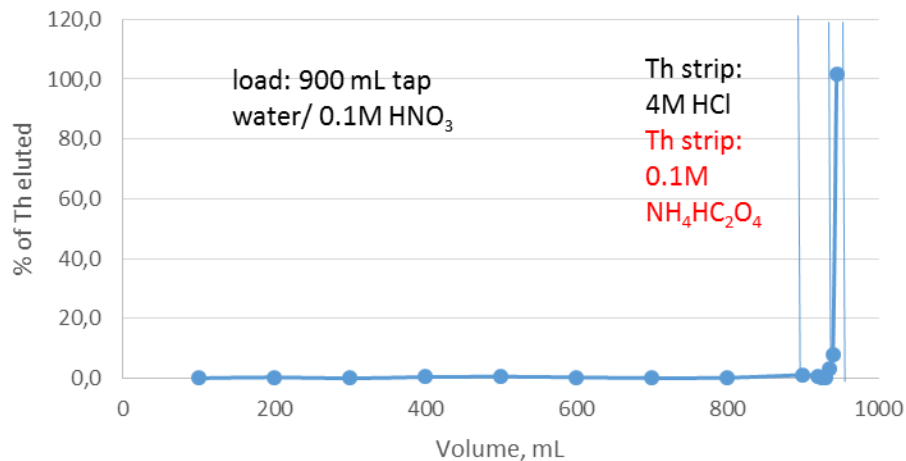


Zhao Huang, Xiaolin Hou, Xue Zhao, Rapid and Simultaneous Determination of ²³⁸Pu, ²³⁹Pu, ²⁴⁰Pu, and ²⁴¹Pu in Samples with High-Level Uranium Using ICP-MS/MS and Extraction Chromatography, Anal. Chem. 2023, 95, 34, 12931–12939, <https://doi.org/10.1021/acs.analchem.3c02526>

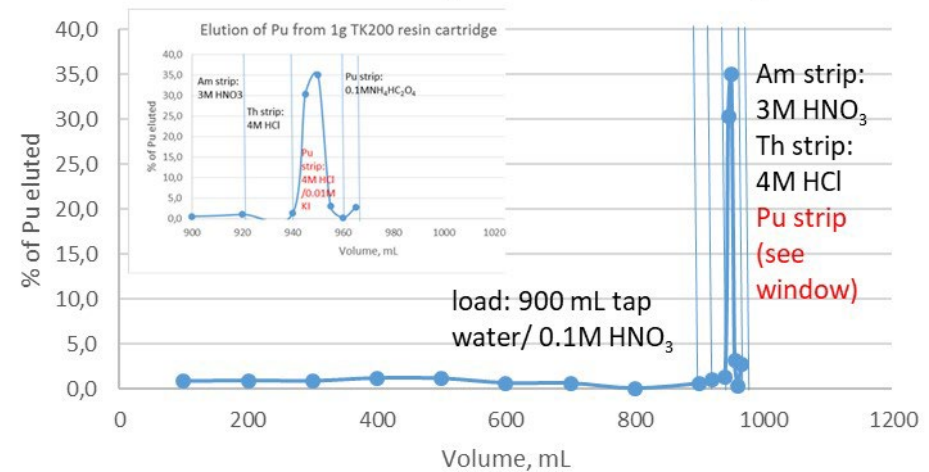
Figure S1. Analytical procedure for determination of plutonium isotopes (²³⁸Pu, ²³⁹Pu, ²⁴⁰Pu, ²⁴¹Pu)

Actinides on TK200 – Preconcentration

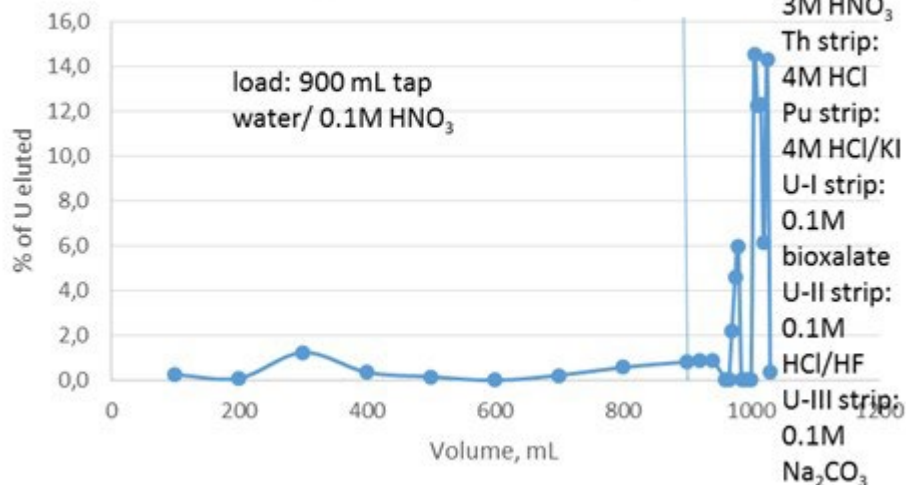
Elution of Th from 1g TK200 resin cartridge



Elution of Pu from 1g TK200 resin cartridge



Elution of U from 1g TK200 resin cartridge



- Data by N. Vajda (RadAnal)
- Method development on-going
- Direct load of U, Th and Pu from acidified water samples (here 900 mL on 1g TK200)
 - ~10 mL/min)
- Sequential separation on TK200
- Automatisations & 'in the field' preconcentration?

TK200 – direct Pu load/separation



- Acidified water samples (1 L)
- One TK200 cartridge
 - Preconcentration and purification
- Automated separation
- Flow rate 15 mL/min
- $D_f(U)$: $10^4 - 10^5$
- LoD:
 - 0.32 $\mu\text{Bq/L}$ Pu-239
 - 2.00 $\mu\text{Bq/L}$ Pu-240

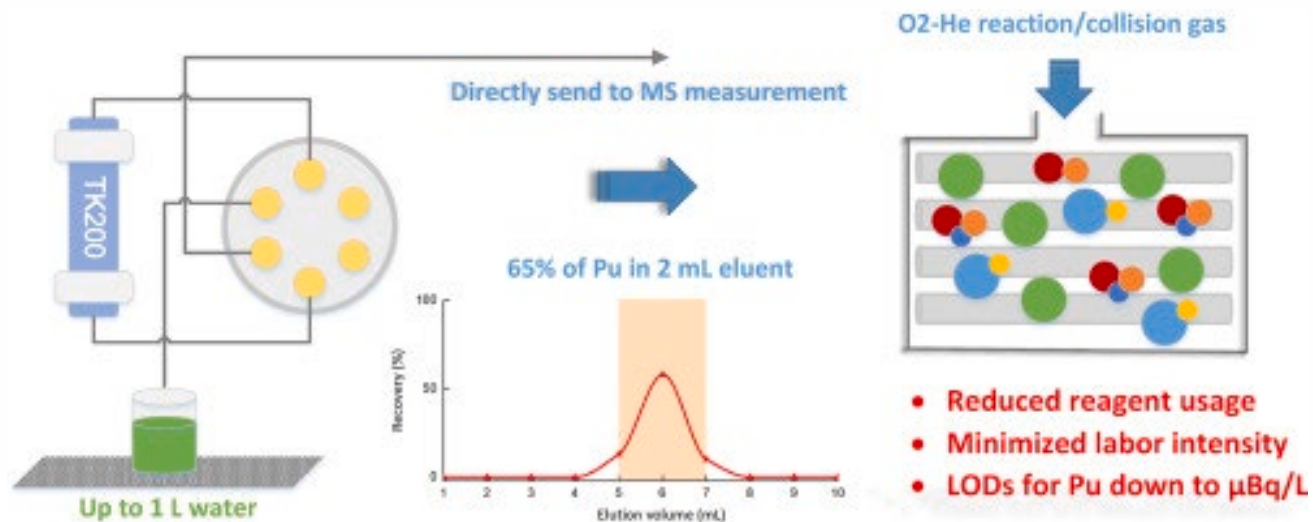


Talanta
Volume 262, 1 September 2023, 124710



A novel strategy for Pu determination in water samples by automated separation in combination with direct ICP-MS/MS measurement

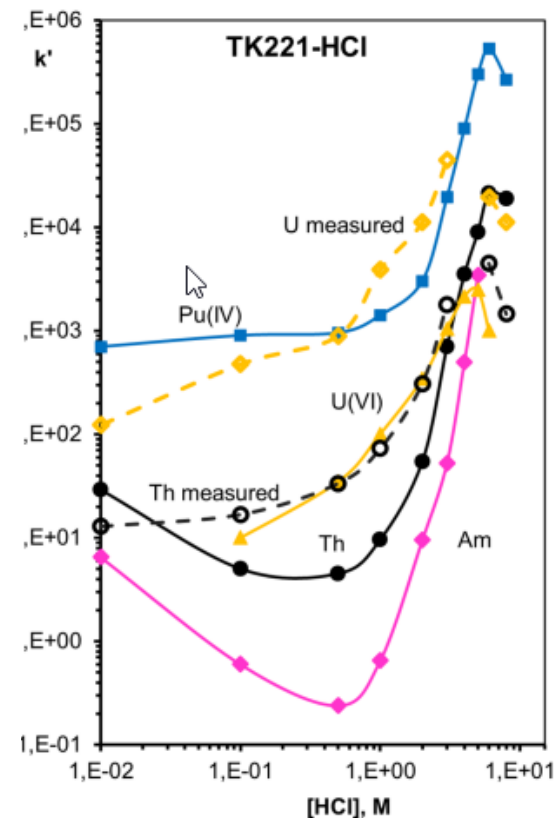
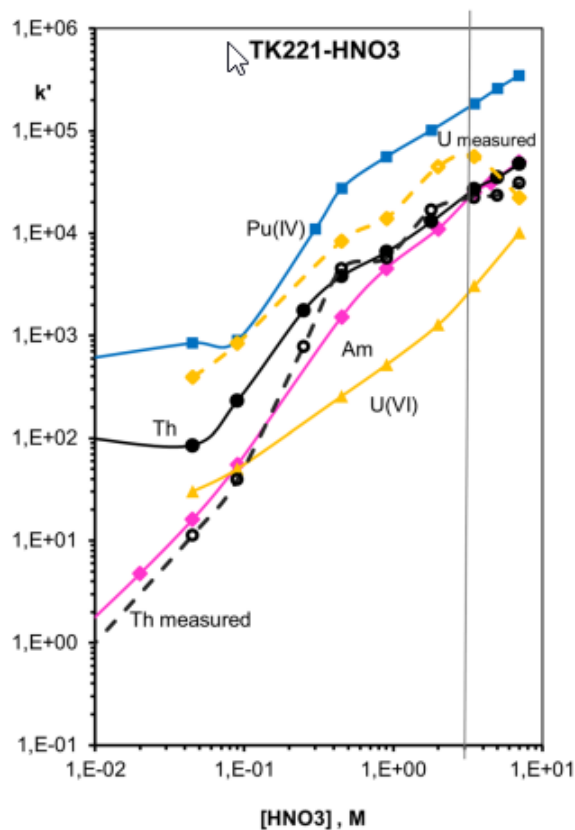
Yuyi Ni^a  , Wenting Bu^a, Ke Xiong^a, Sheng Hu^a, Chuting Yang^a, Liguao Cao^b



'New' - TK221 Resin

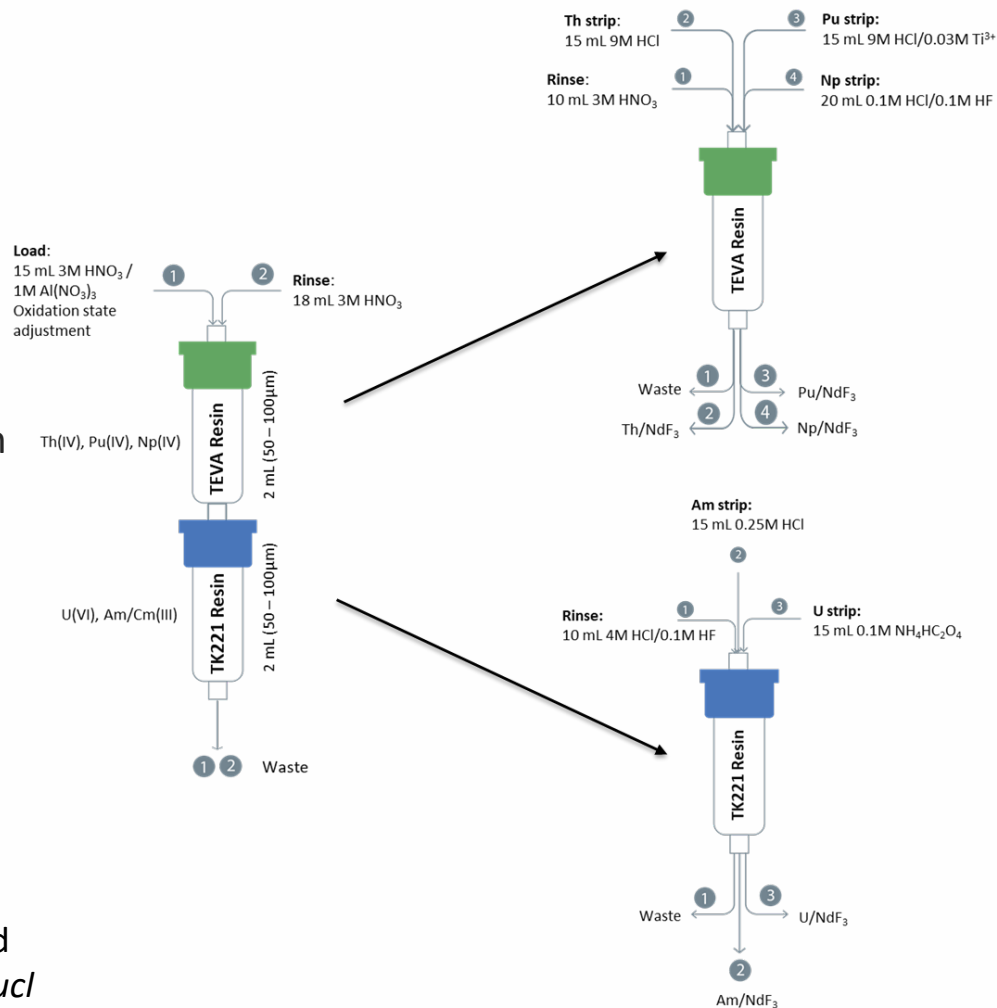
Resin based on a mixture of diglycolamide and phosphine oxide + traces long chained alcohol on inert support.

- Main applications in RadPharm:
 - Lu, Tb concentration from high acid and elution in small volume of dilute HCl
 - Ac-225 purification
- Potential interest for Actinides separation
 - Higher U retention than DGA
 - Higher Am retention than TRU



TK221 Resin – actinide separation

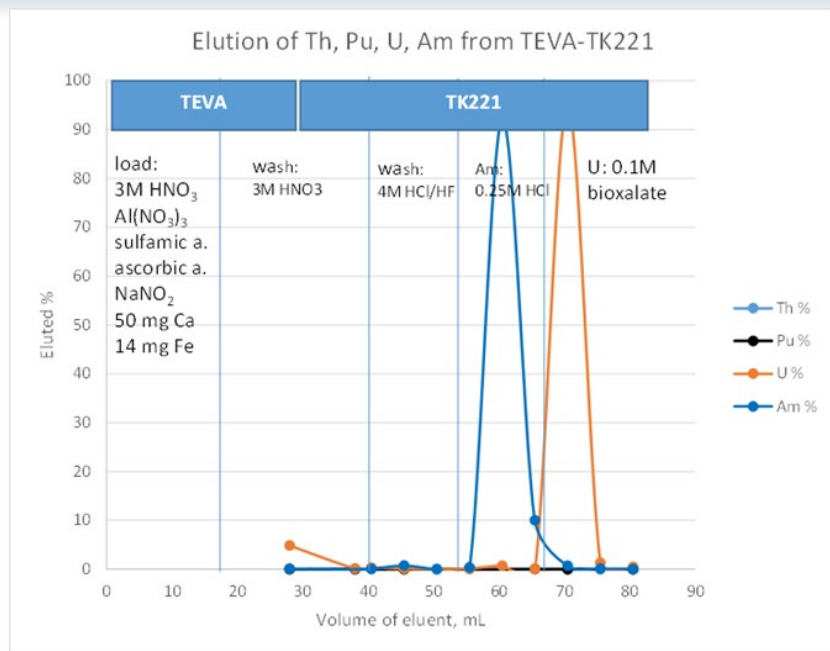
- Cooperation with Nora Vajda
- Method development for water samples
- Ca Phosphate co-precipitation
- TEVA/TK221 separation method
 - TEVA: Pu(IV)/Np(IV) and Th(IV) separation
 - TK221: U(VI) and Am(III) separation
- Ideally later also test on soil and decommissioning samples



Papp, I., Vajda, N. & Happel, S., An improved rapid method for the determination of actinides in water. *J Radioanal Nucl Chem* **331**, 3835–3846 (2022).

<https://doi.org/10.1007/s10967-022-08389-9>

TK221 Resin



Th and Pu removed with TEVA.

Am and U separation on TK221 Am elution before U

Table 3 Recovery of actinide tracers from spiked water samples

	Actinides determination	
	Without Np separation	With Np separation
	Yield %	Yield %
TAP water		
²³⁰ Th	90 ± 8	86 ± 7
²³⁹ Pu	108 ± 7	95 ± 7
²³⁷ Np	—	91 ± 9
²⁴¹ Am	103 ± 7	97 ± 6
²³³ U	103 ± 7	70 ± 7
SEA water		
²³⁰ Th	71 ± 7	61 ± 6
²³⁹ Pu	91 ± 7	87 ± 6
²³⁷ Np	—	93 ± 8
²⁴¹ Am	89 ± 7	92 ± 6
²³³ U	88 ± 7	59 ± 6

Analyte	Target values		Measured values			Relative bias %	MARB ^a %	Z-score ^b	Test evaluation
	Mean activity concentration	Standard deviation (sd)	Activity concentration	Standard uncertainty	Relative standard uncertainty				
	Bq/kg	Bq/kg	Bq/kg	Bq/kg	%				
²³⁹ Pu	5.93	2.27	5.09	0.24	4.7	14	25	0.37	Accepted
²⁴¹ Am	4.85	0.57	4.73	0.15	3.2	2.5	30	0.21	Accepted
²⁴⁴ Cm	7.02	2	7.19	0.34	4.7	2.4	25	0.09	Accepted

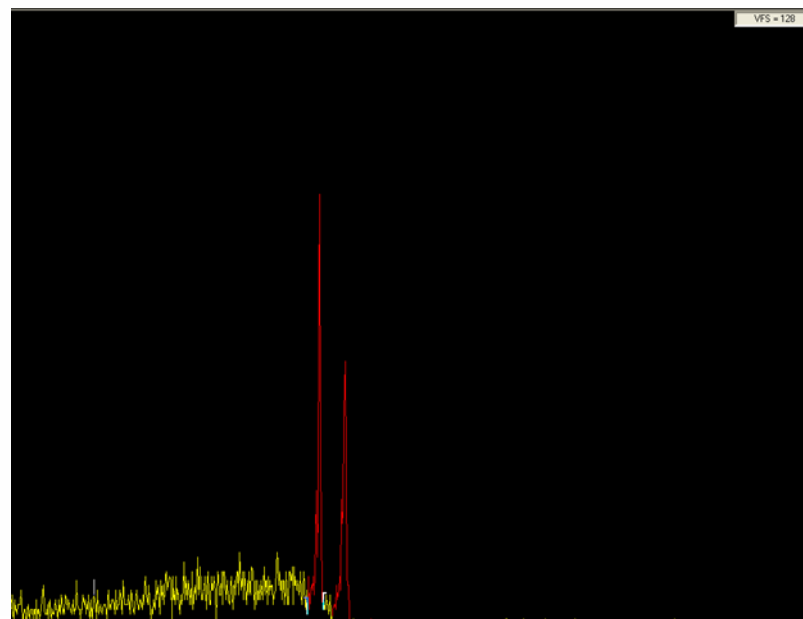
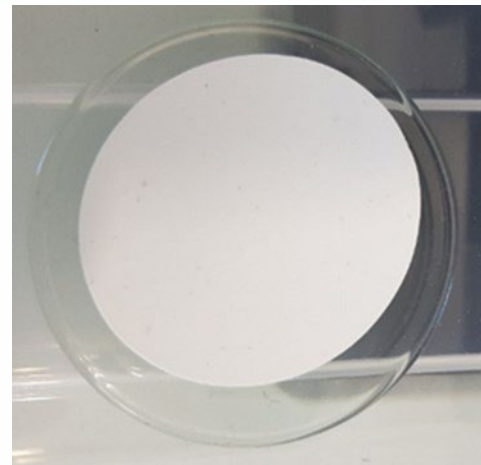
^aMaximum Acceptable Relative Bias

^bZ = |X_{reported} - X_{target}| / s_{dtarget}

- Method tested on spiked tap and sea water samples
- High yields (88+% for U and Am)
- Analysis of IAEA-TEL-2021-03 WWOPT successful
- Next: use for solid samples?

Under development: gross alpha discs

- On-going work: development of impregnated membrane filters
- First filters under beta testing:
 - TK100 (DGT of Sr, Pb, Zn, LN in soil samples)
 - TK201 (determination of Tc-99 in aqueous samples)
 - 25mm and 47mm
- Currently under testing, membrane filter for gross alpha measurement
- pH 2, 10mL/min, typically 100mL samples
- High retention of actinides
- Peak resolution/spectrum still to be improved



Alpha spectrum, Am-241 & Pu-239, ~50mBq each



Thank you for your attention!

In case of questions or interest in R&D cooperation:

shappel@triskem.fr



 **SUBSCRIBE TO OUR NEWSLETTER**
To keep updated with our latest developments, news and agenda for a year, subscribe to the Triskem Infos [here](#)

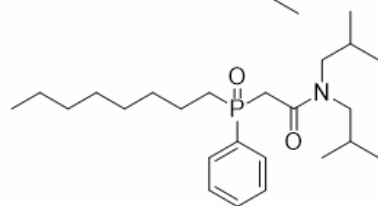
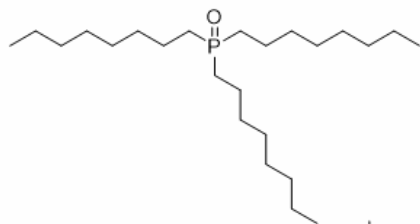


Some other on-going projects

- Impregnated membrane filters
 - AC Discs for gross alpha?
 - Replacement of Nucfilm U discs
 - Passive sampling
- Impregnated PSm resins
- Range of 'Test sticks'
 - Suitable impregnated support
 - JCU => rapide isotope ratio analysis by MS (metallomics)
 - Uni Southampton/NPL
 - Ideally multiple layers of resins for multi RN screening
 - LSC measurement
 - Scintillating supports for non-LSC options
 - Decommissioning/screening
- Separation of DTM
 - SE Resin
 - Zr-93, Fe, Mo, Nb,...
- Fate' of RN in the environment
 - Separation methods
 - Mainly longer lived RN (=> therapy)
 - Ac-225/7, Lu-177(m), radioiodine,...
 - Quantification
- In-field preconcentration
 - Impregnated membranes
 - Cartridges
- Microfluidics
- Other 'geometries' & 'Non-resin' separation materials

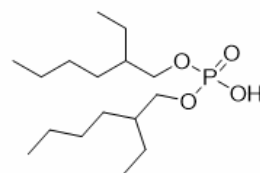
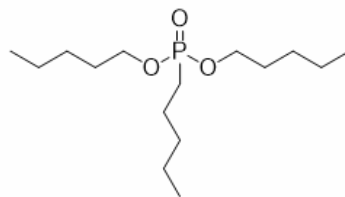
Typically employed extractants

**Trioctylphosphine oxide
(TOPO)**



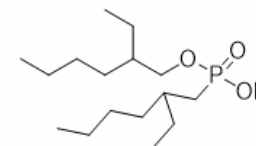
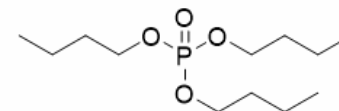
**Octyl (phenyl)-N,N-diisobutyl-
carbamoylmethylphosphine oxide
(CMPO)**

**Dipentyl pentyl phosphonate
(DPPP)**

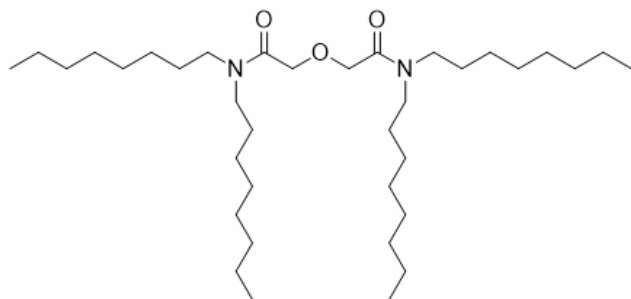


**Bis(2-ethylhexyl) hydrogen
phosphate
(HDEHP)**

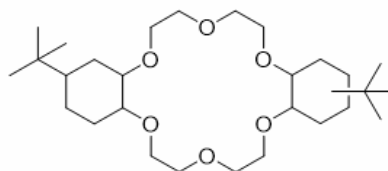
**Tributylphosphate
(TBP)**



**2-Ethylhexylphosphonic acid
mono-2-ethylhexyl ester
(HEH[EHP])**



**N,N,N',N'-tetra-n-octyl-diglycolamide
(TODGA)**



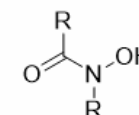
**4,4'(5')-di-t-butylcyclohexano-18-
crown-6**

Aliquat 336 (R = 8, 10)



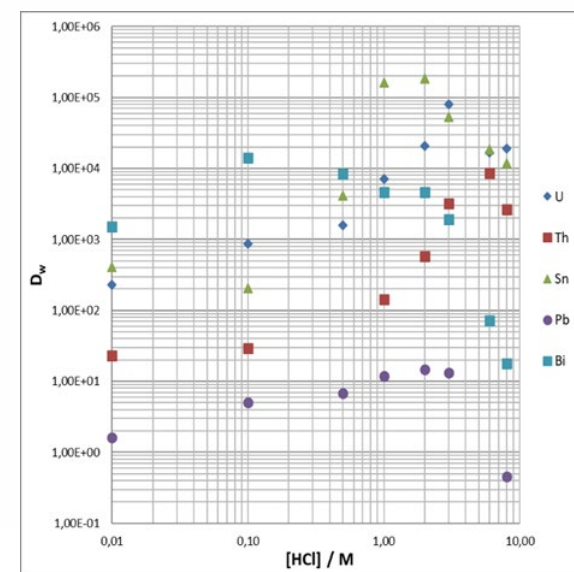
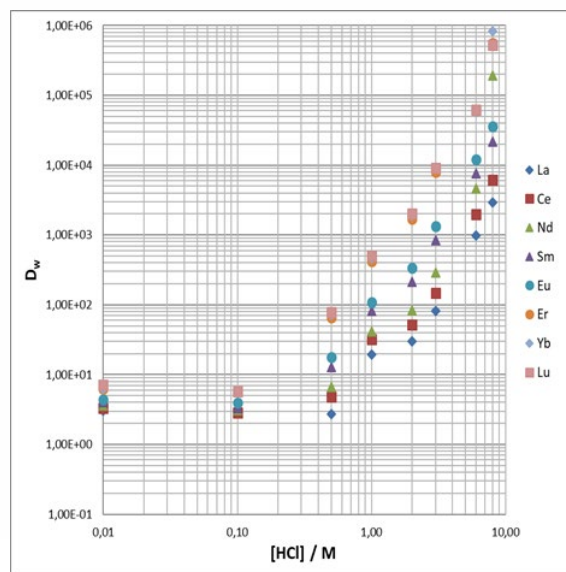
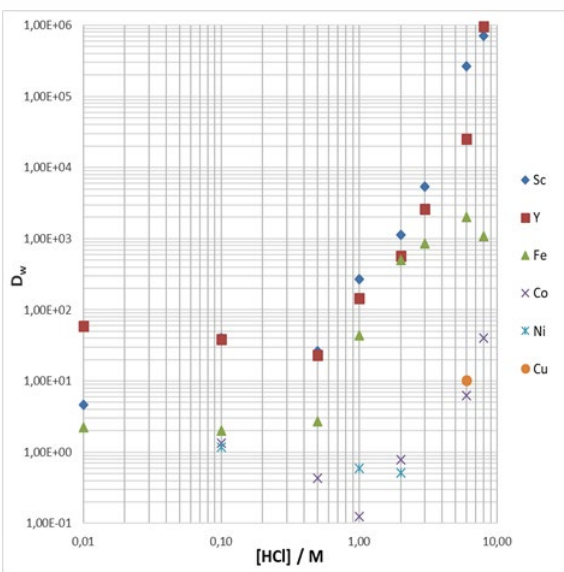
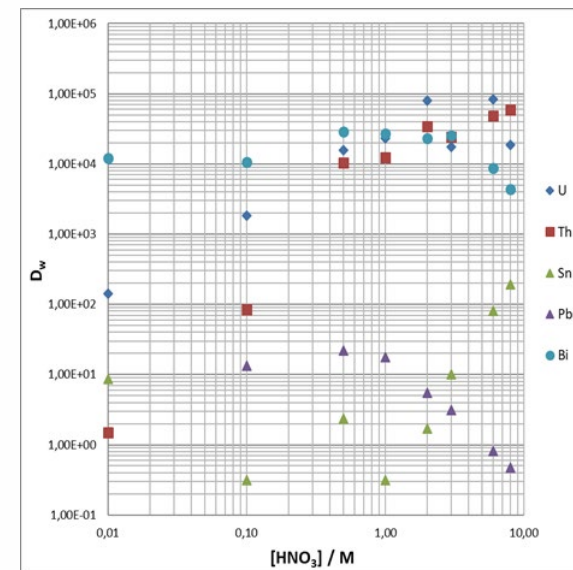
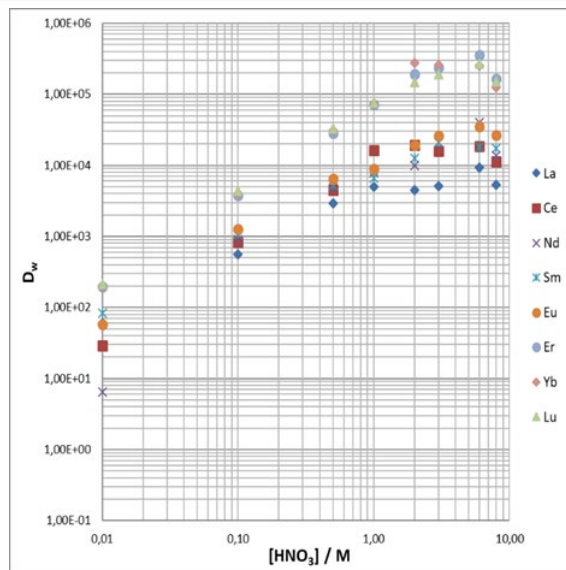
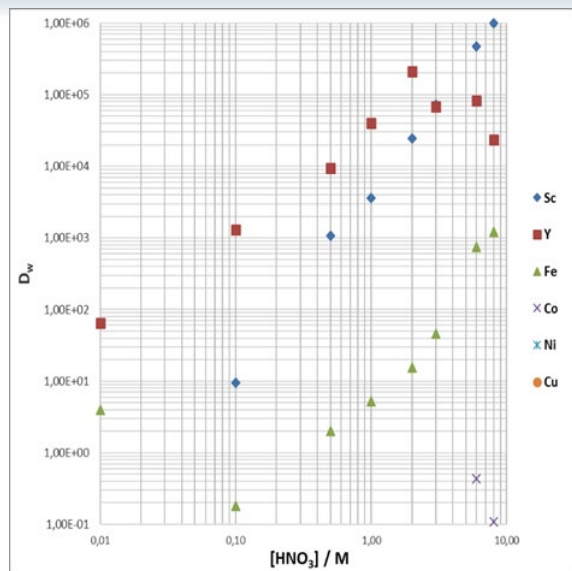
Phosphine sulfide

Tertiary amine



Hydroxamate

TK221 Resin => product sheet...



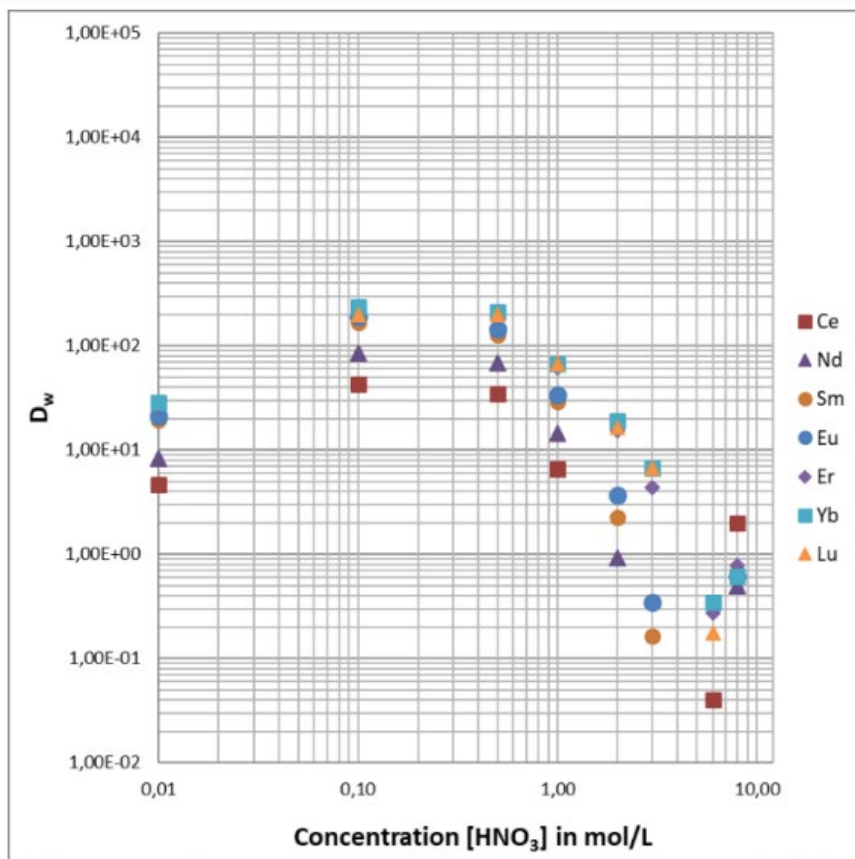


Figure 6: D_w values of selected elements on TK200 Resin in HNO_3