



TrisKem International

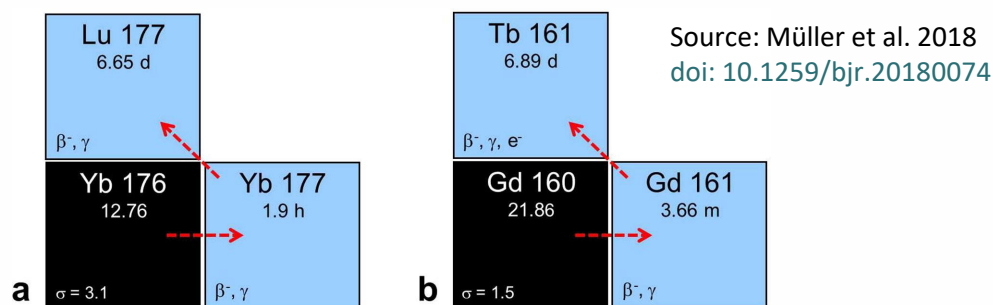
On the development of a method for the separation of
Terbium from elevated amounts of Gadolinium using TK221
and TK211/2 Resins

Illarion Doyhi, Anaëlle Vilboux, Marine Bas, Aude Bombard, [Steffen Happel](#)

25/06/2024



- nca Lu-177 very widely used but Tb-161 getting strong interest
 - Part of the ‘Swiss knife of nuclear medicine’ => Tb isotopes
- Similar n.c.a. production for both



Tb 149		Tb 152		Tb 155	Tb 161
4.2m	4.1 h	4.2m	17.5h	5.32 d	6.90 d
ϵ	ϵ	γ 283;	ϵ	ϵ	β^- 0.5; 0.6...
β^+	α 3.97	160...	β^+ 2.8...	γ 87;	γ 26; 49; 75...
α 3.99	β^+ 1.8	$\epsilon; \beta^+$...	γ 344;	105;...	e^-
γ 796;	γ 352;	γ 344;	586;	180, 262	
165...	165...	411...	271...		

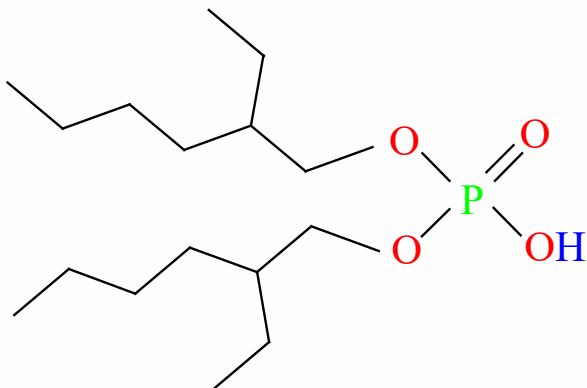
Terbium: a new ‘Swiss army knife’ for nuclear medicine

Source: <https://cerncourier.com/a/terbium-a-new-swiss-army-knife-for-nuclear-medicine/>

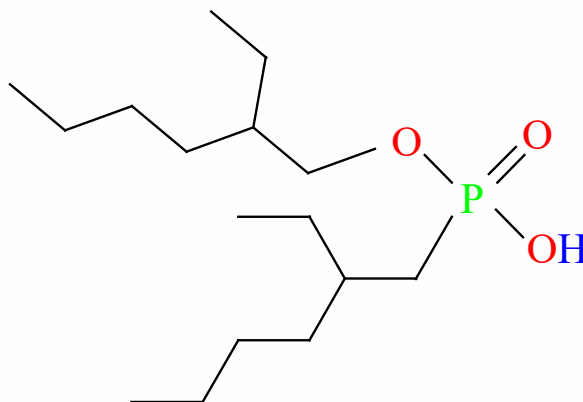
- Irradiation of several hundreds of mg or more
- Upscale on-going (incl. recycling/waste treatment) => typically ≥ 1 g
- Tb very similar to Lu chemically and with respect to half-life et al.
- Additionally emits significant number of Auger-Meitner electrons
- Availability of Gd-160 problematic

- Increasing demands for separation from larger Yb, Gd,... targets
- Resins exposed to high radiation throughout separation process
- Desire to re-use columns several times
 - Improvement of radiolysis stability
- Feedback from earlier project => stability against radiolysis can be improved via:
 - Use of polymer containing aromatic groups as inert support
 - Addition of radical scavenger (e.g. long chained alcohols) into stationary phase
 - Increased amount of extractant and nature of extractant
 - EtOH in aqueous phase
- Applied to TK211/2/3, TK221/2
- Note: more resins are more hydrophobic, require soaking in $\geq 20\%$ EtOH for column packing

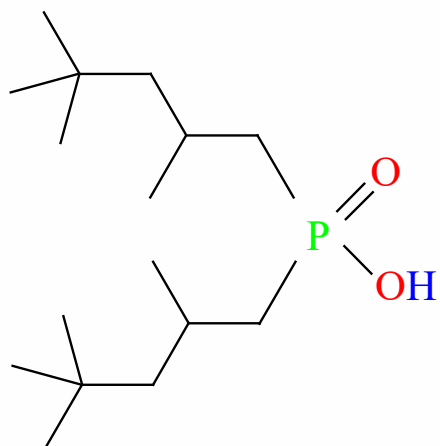
Lanthanide separation on TK211/2/3



HDEHP (LN)

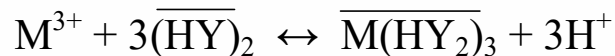


HEH[EHP] (LN2)



H[TMPeP] (LN3)

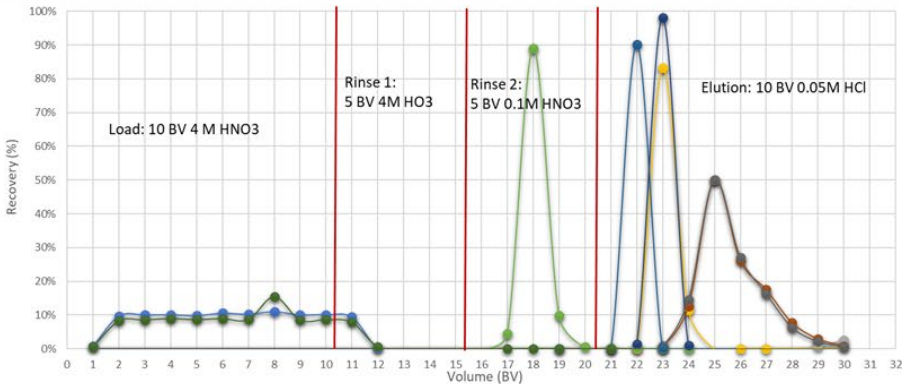
Cyanex 572



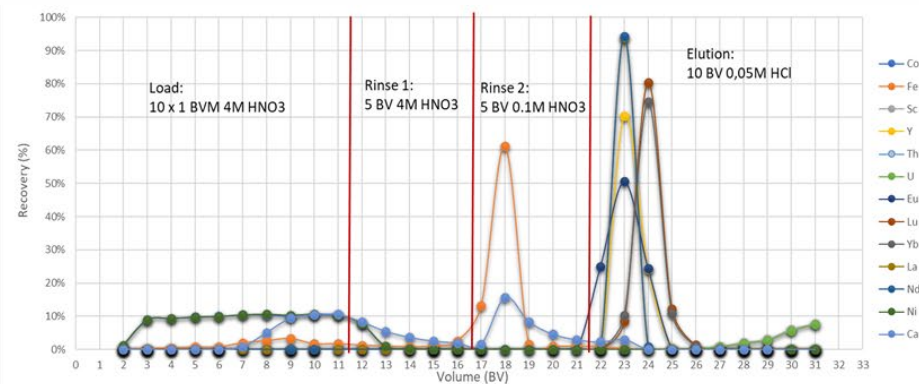
- Mixtures of different extractants
- Optimized for high radiation stability

TK221 Resin

- DGA well suited for ‘conversion’ and purification (Ca, Al, Fe,... removal)
 - Convert Lu from high nitric acid to dilute HCl => feedback, quicker Lu elution desired
- TK221 Resin
 - DGA / phosphine-oxide based => phosphine oxide should improve radiolysis stability
 - Better La and U, Th retention than DGA
 - Lu & Tb eluted in small volume in dilute HCl
 - Drawback, no (group) LN separation possible!
- TK221 also used for Tb recycling
 - Loading ‘Gd recycling fraction’ onto TK221
 - Elution in 0.05M HCl => suitable conditions for oxalate precipitation

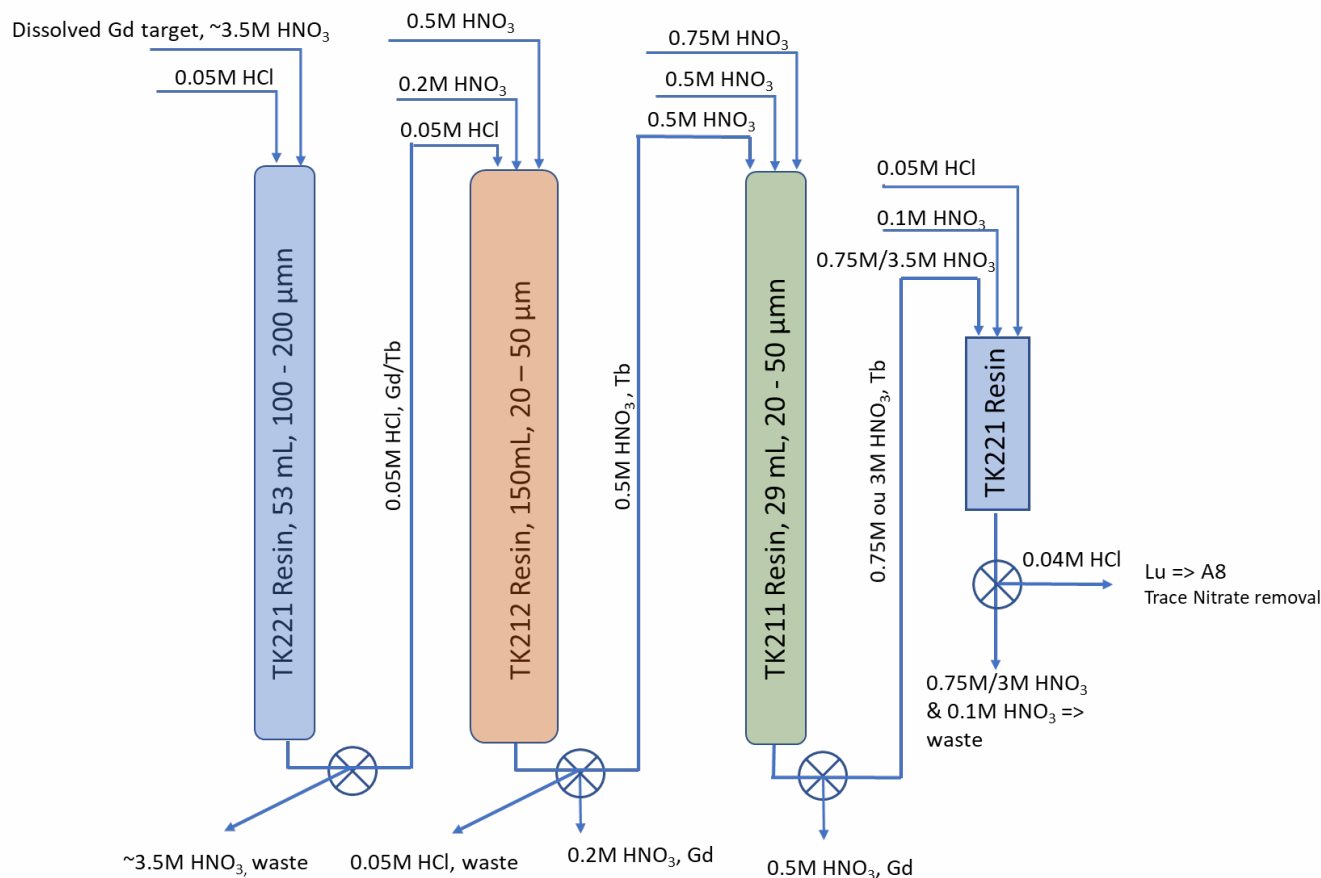


DGA Resin



TK221 Resin

Separation method

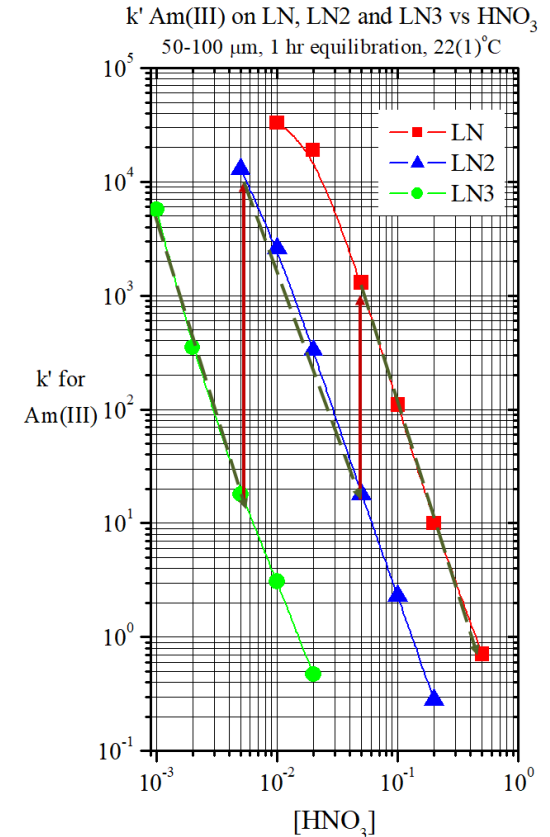
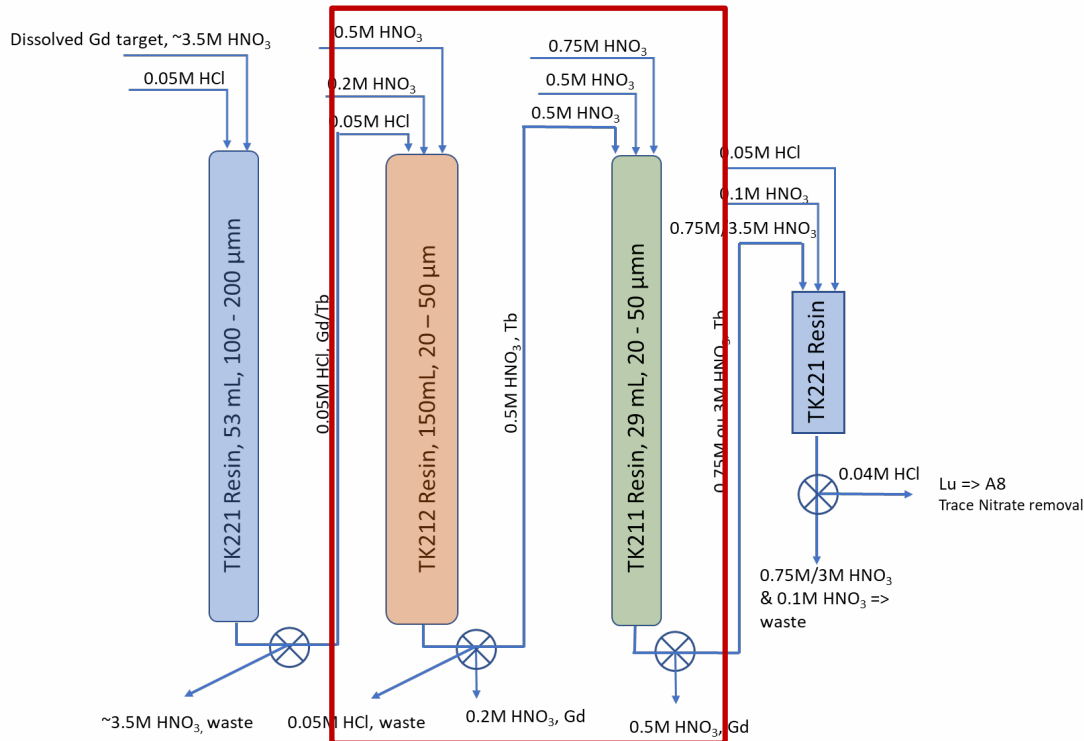


- TK221, TK212/1 based method
- First TK221 column size depending on amount of Gd
- For very large Gd amounts TK212 column might need to be adjusted, too

- Work with pre-packed PP columns (PEEK also possible)
 - ¼"28G connectors
- Focus on TK212/TK211
 - TK212: 30cm x 2.5cm column
 - For lower amounts of Gd (e.g. $\leq 100\text{mg}$) smaller TK212 columns may be used
 - TK211: 30cm x 1.1cm column
 - First TK221 needs to be adapted to amount of Gd
- HPLC driven separations (10 – 15 mL/min)
- Cold testing by taking fractions of defined volume
 - Typically way more fractions than needed, helpful for understanding the chromatographic separation
- ICP-MS analysis of fractions



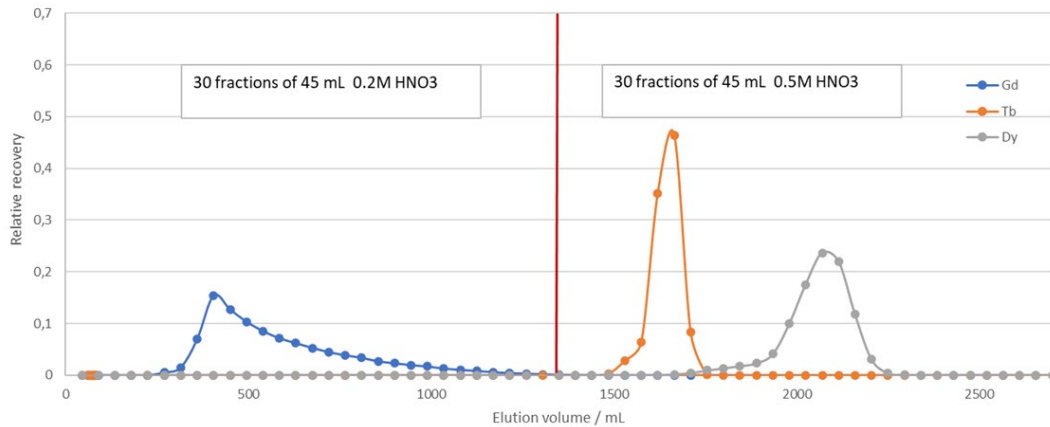
Sequential separation step



- Feedback: while conversion from high acid via TK221 or DGA works well, users frequently report losses of up to 2 - 5%
- Directly loading from TK212 onto TK211 avoids this conversion step and associated losses. Draw back => increased pressure drop

Initial work: 500mg and 1000mg Gd

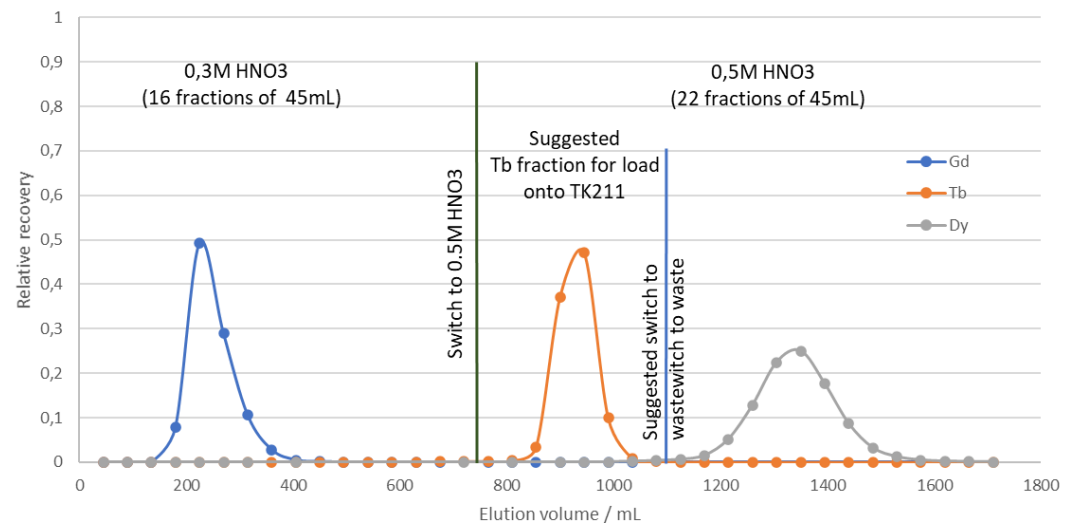
TK212 (150 mL) - 500 mg Gd / 500 µg Tb / 500 µg Dy - 15 mL/min



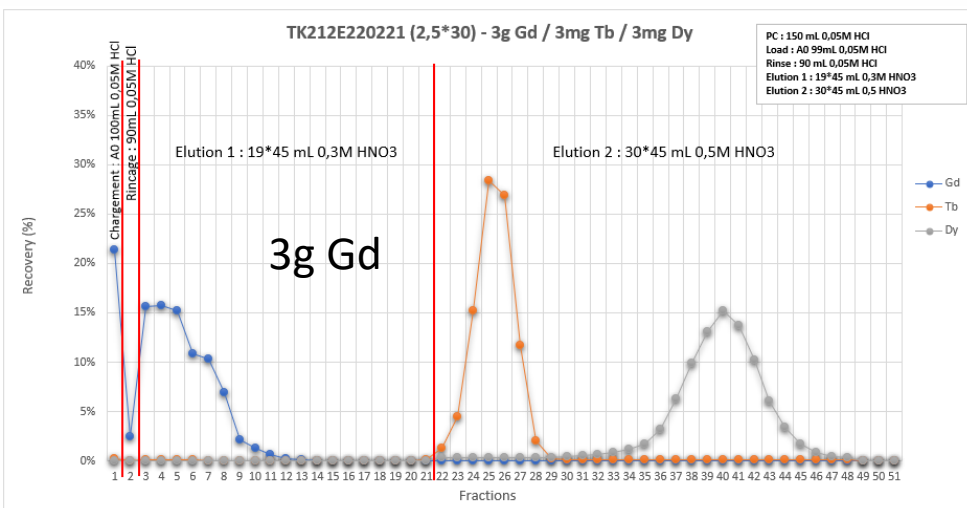
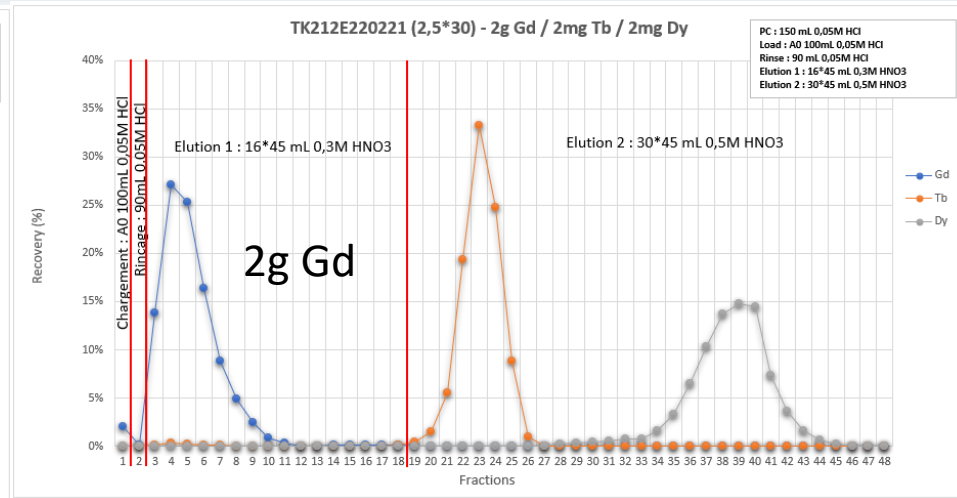
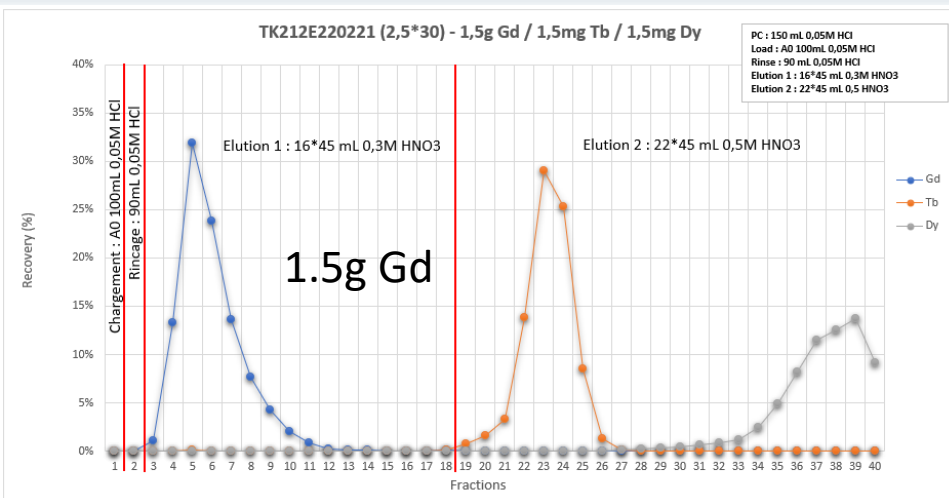
- Initial test with 500 mg Gd
- Gd removal with 0.2M HNO₃
- Strong tailing
- Still very good separation

- Eliminating Gd with 0.3M HNO₃
- Larger amount of Gd (1g)
- Less tailing
- Clean separation

TK212 (150 mL) - 1g Gd / 1mg Tb / 1 mg Dy - 15mL/min

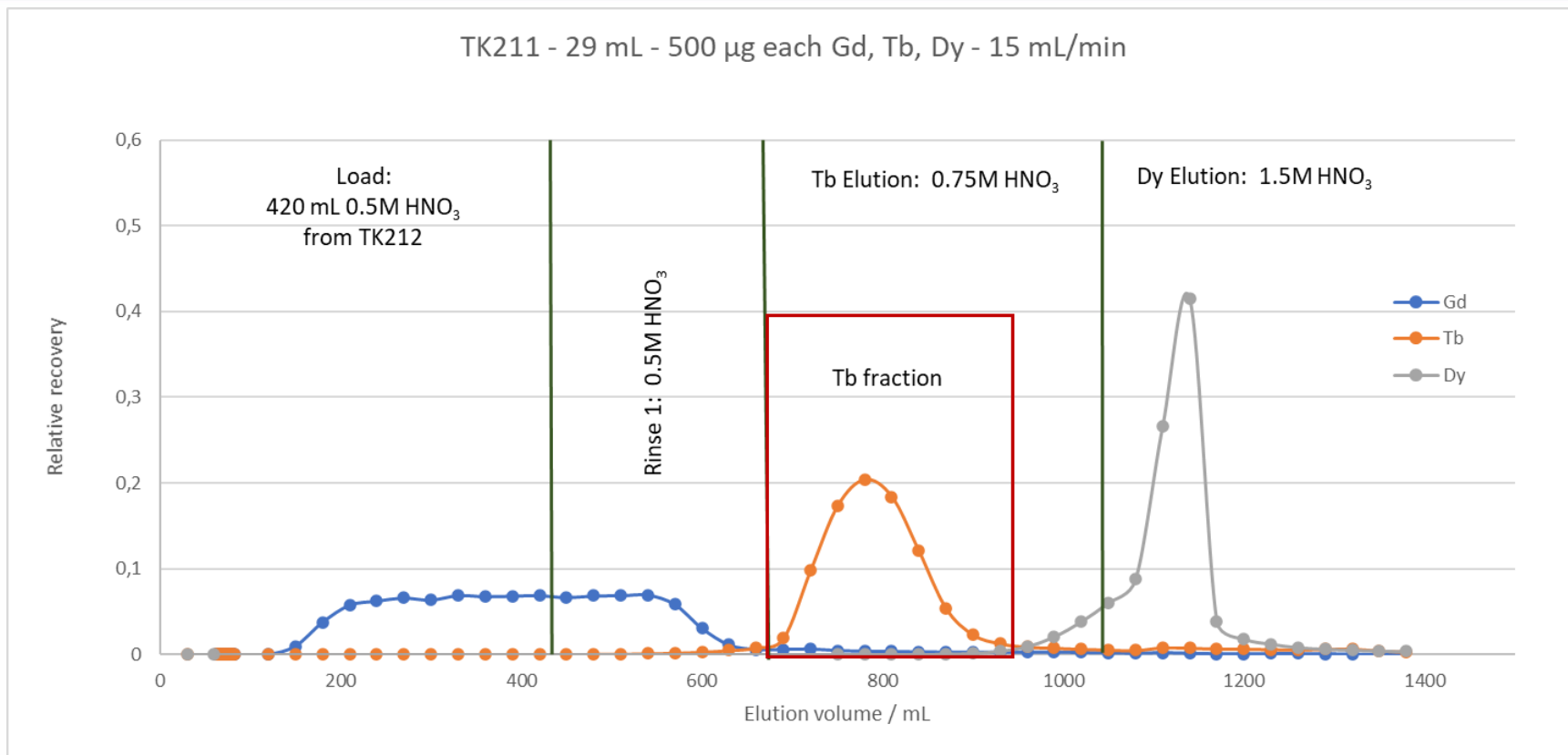


Increased amounts of Gd



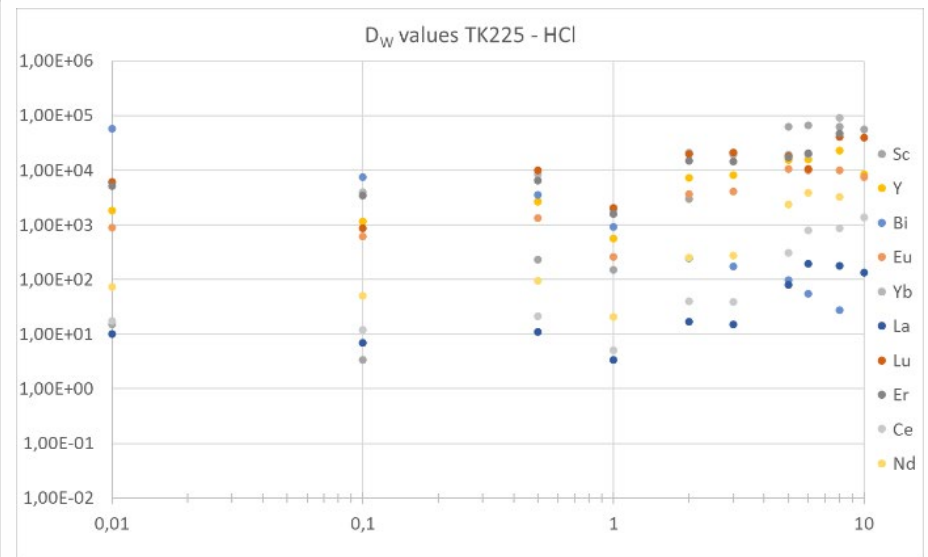
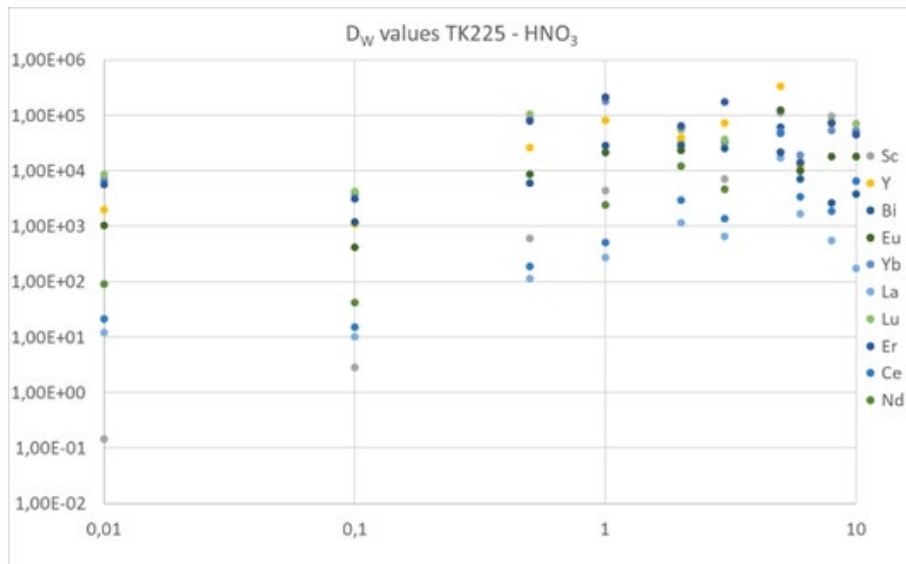
- On the same TK212 column
- More Gd => earlier elution
 - At 3g Gd start of breakthrough
 - More than 3g possible? Tb needs to remain retained...
- Little effect on Tb
- Small impact on Dy
- Tb / Dy separation remains good.

Tb polishing on TK211



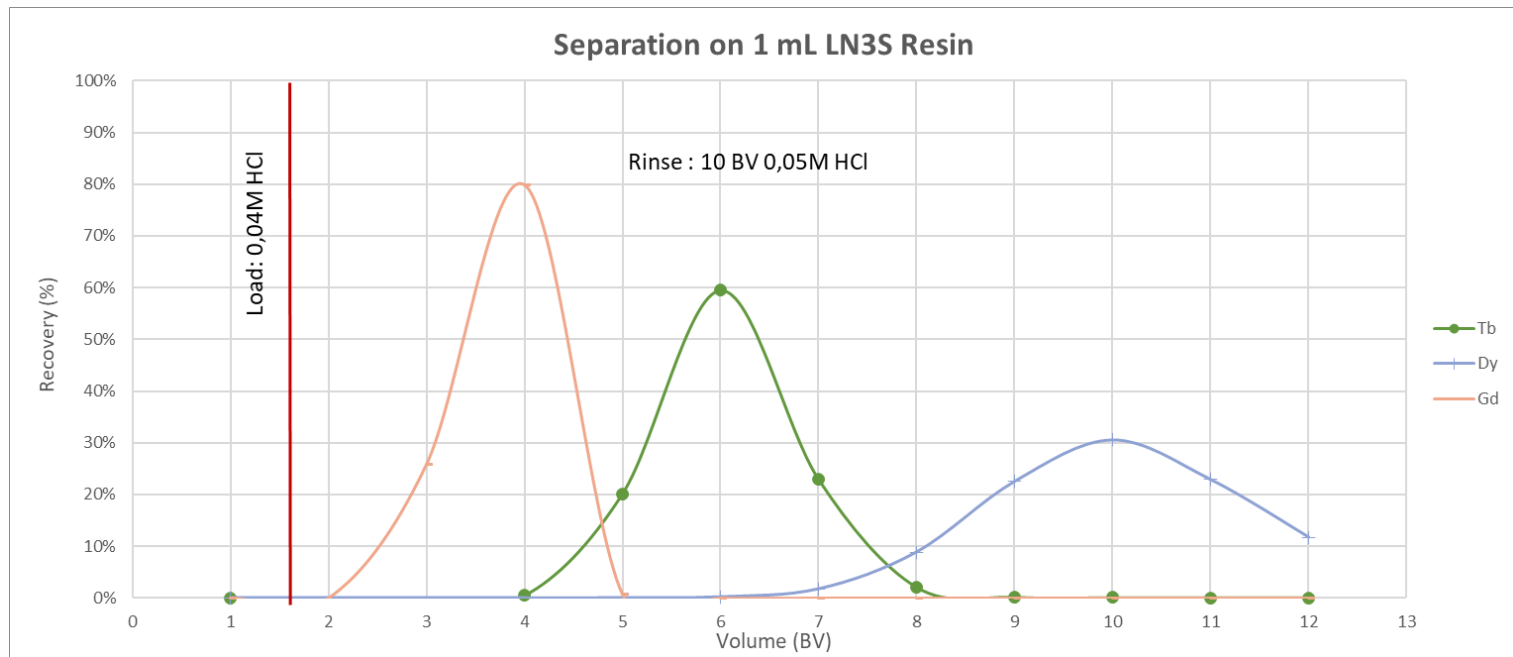
- Spiked solution to allow for obtaining a chromatogram
- Direct load of Tb fraction from TK212 onto TK211 (29 mL – 30cm x 1.1cm)
- Gd breakthrough during load & rinse with 0.5M HNO₃ (alternatively HCl)
- Feedback: Tb elution (Dy sufficiently well removed on TK212) in **>3M HNO₃**
- Conversion to dilute HCl via 2 mL TK221, A8 for nitrate removal

- TO-DGA plus ionic liquid
- High retention of light lanthanides at medium to high acid
- Heavy lanthanides also very well retained at low acid concentrations
- **Main application: Removal of radiolanthanides from effluents**



On-going: Tb/Dy separation

- Request:
 - Removal of Dy ingrown e.g. during transport
 - Aim: rapid removal, no significant change of volume and acidity of Tb
 - On-going work



- Further up-scale
 - Same TK212 column (30cm x 2.5cm): 5g Gd,...
 - Gd will increasingly breakthrough during load
 - Pro:
 - Gd not fully retained during loading not problematic as long as Tb remains retained
 - May allow for treating very large targets
 - Contra:
 - Co-elution of potentially present impurities/no purification of the Gd (Eu...)
 - Use of 4 cm x 30cm TK212 for very large amounts of Gd
- Integrate Eu removal step into TK212 separation if needed
- Er/Tm separation (currently at 1g Er)
- Further Yb/Lu Upscale



Thank you for your attention!



SUBSCRIBE TO OUR NEWSLETTER

To keep updated with our latest developments, news and agenda for a year, subscribe to the TrisKem Info here



Tb separation from 10mg Gd

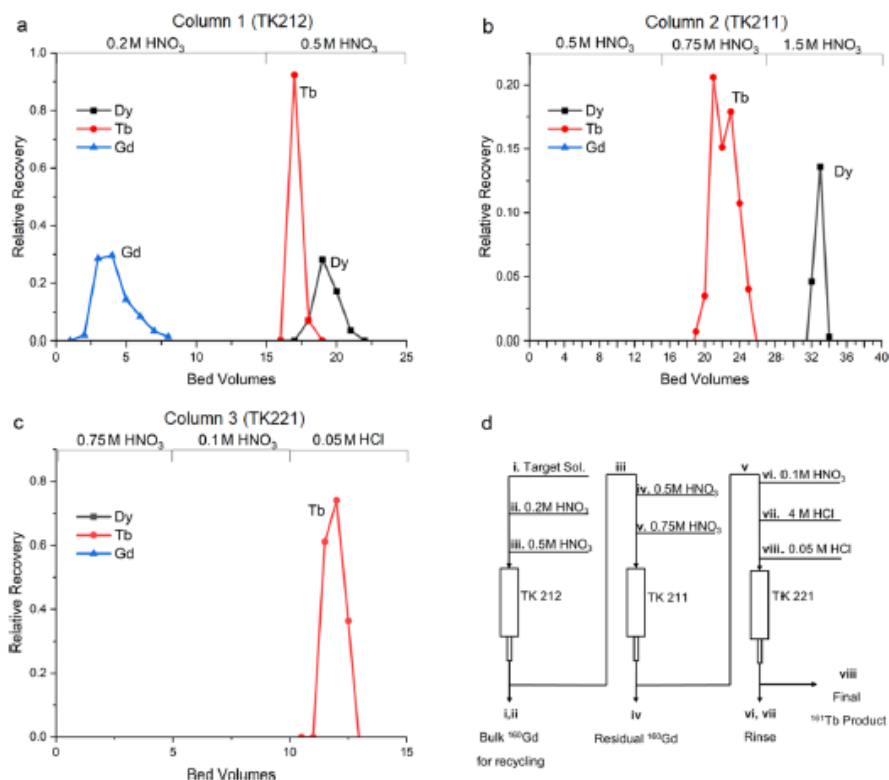


Fig. 3 Elution profile of Gd, Tb, and Dy (10 mg, 1 mg, and 1 mg respectively) TK212 (a), TK211 (b) and TK221 (c). Tb containing fractions were combined for loading on the next column to simulate purification run. Column bed volumes were 1 mL for each column during cold test but were later optimized for active runs. Overall process diagram (d) note steps i–v can be performed automatically, while steps vi–viii must be performed manually

McNeil et al.
EJNMMI Radiopharmacy and Chemistry (2022) 7:31
<https://doi.org/10.1186/s41181-022-00183-y>

EJNMMI Radiopharmacy
and Chemistry

RESEARCH ARTICLE

Open Access

A simple and automated method for ¹⁶¹Tb purification and ICP-MS analysis of ¹⁶¹Tb

Scott W. McNeil¹, Michiel Van de Voorde², Chengcheng Zhang³, Maarten Ooms², François Bénard³, Valery Radchenko^{1,4} and Hua Yang^{1,5*}

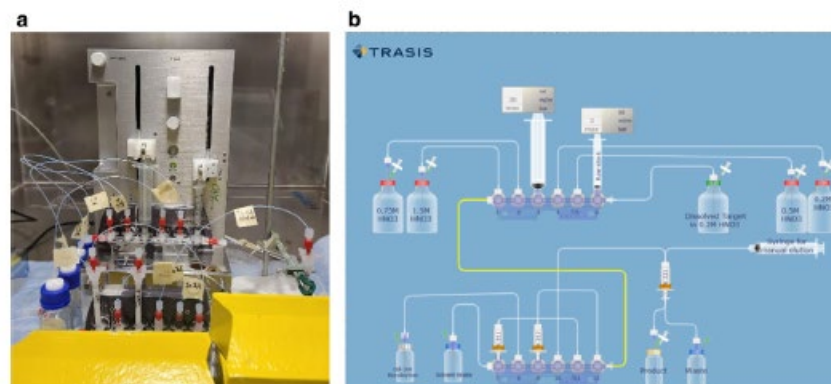
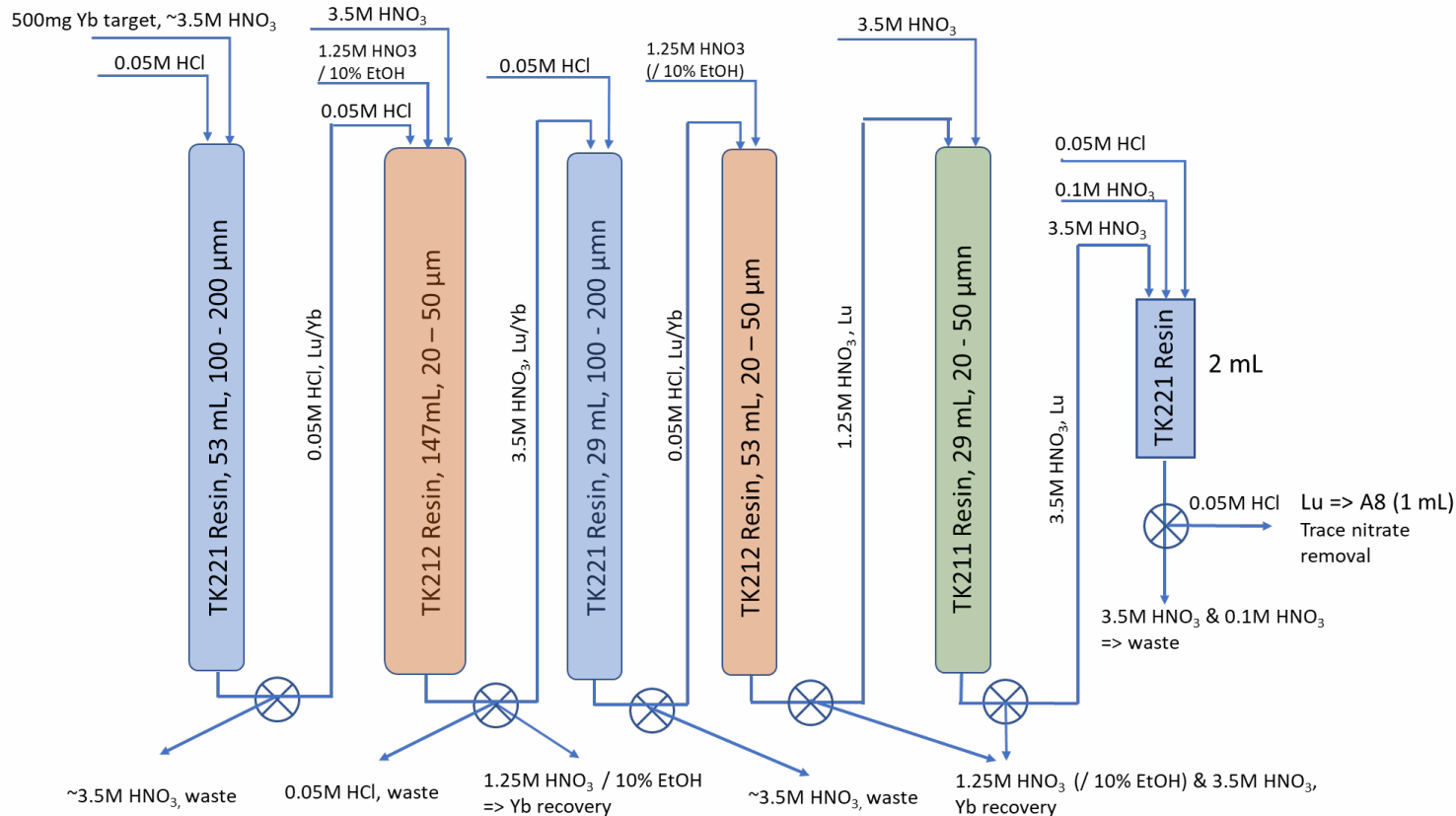


Fig. 4 TRAVIS Mini AIO module and components set-up within hotcell (a), TRAVIS layout diagram (b)

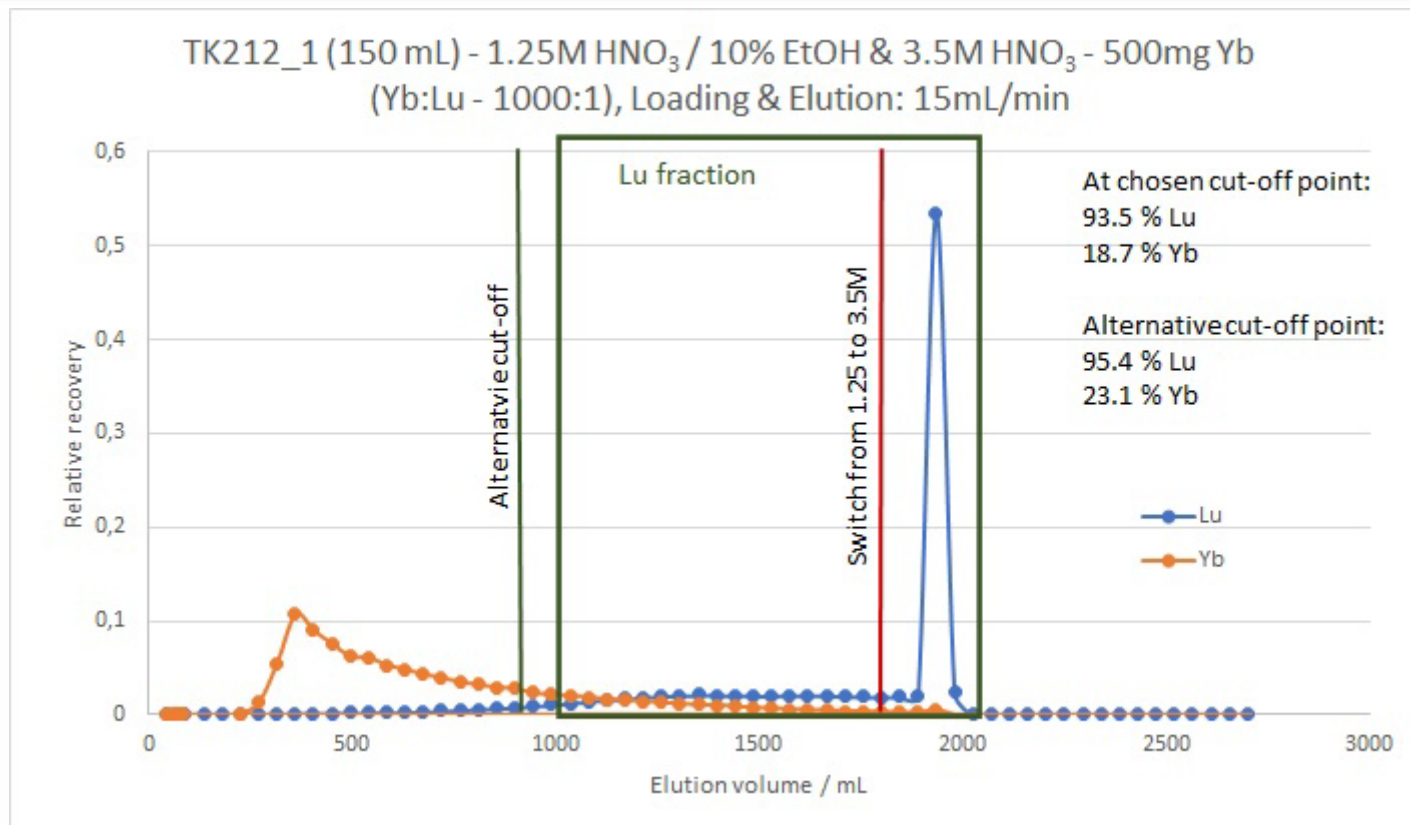
- Fully automated method (synthesizer) for Tb separation from 10mg Gd
- TK212, TK211 and TK221 based system

- Simplified method for Lu separation from 500 mg Yb – TK211/2 & TK221



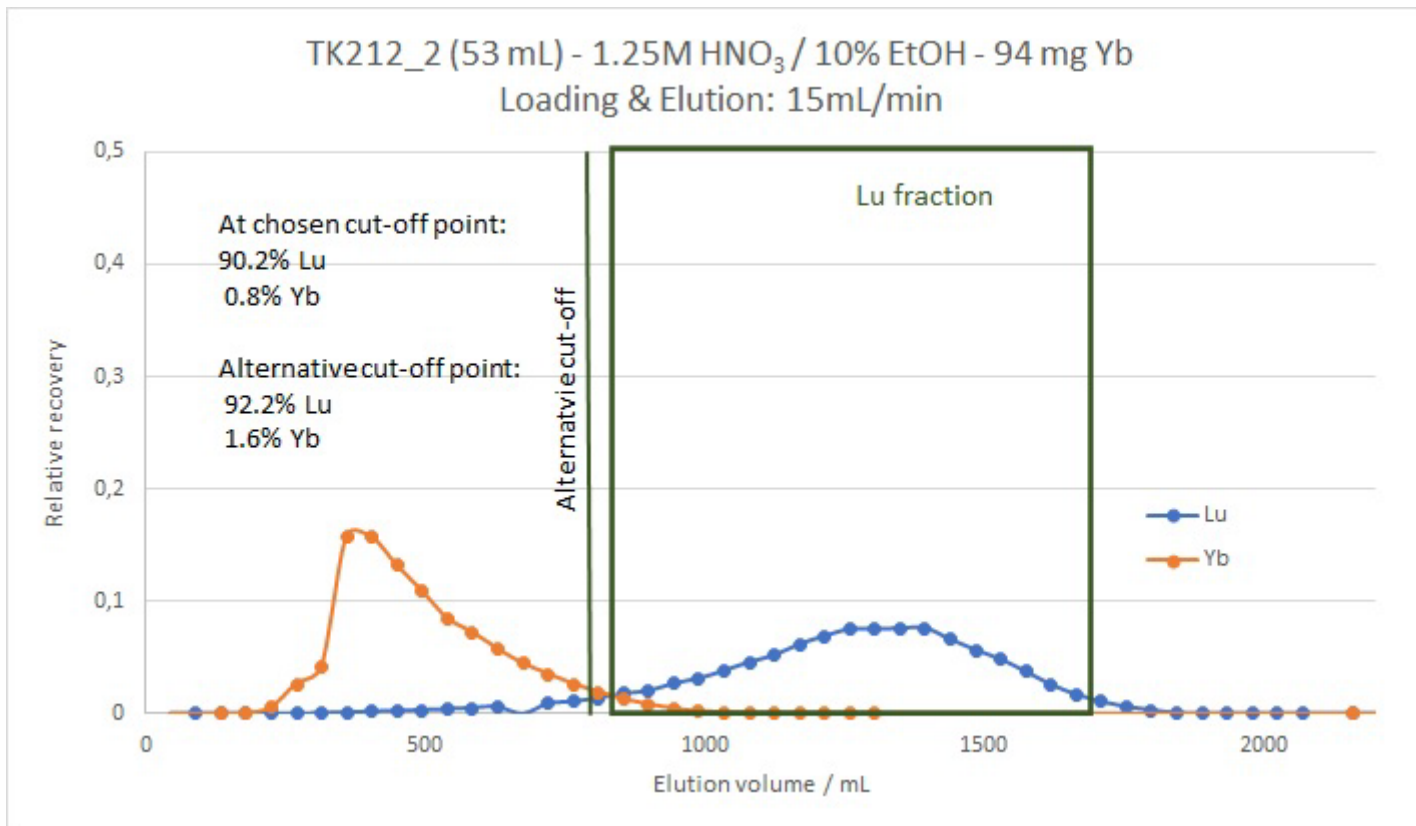
- Sequential separation step (direct load from TK212 onto TK211 for polish)
- Unfortunately complete sequential TK213=>TK212=>TK211 didn't work out
- Can be upscaled (larger columns,...)
- Further optimisation on-going

Lu separation from 500 mg Yb - TK212/TK221/TK212/TK211



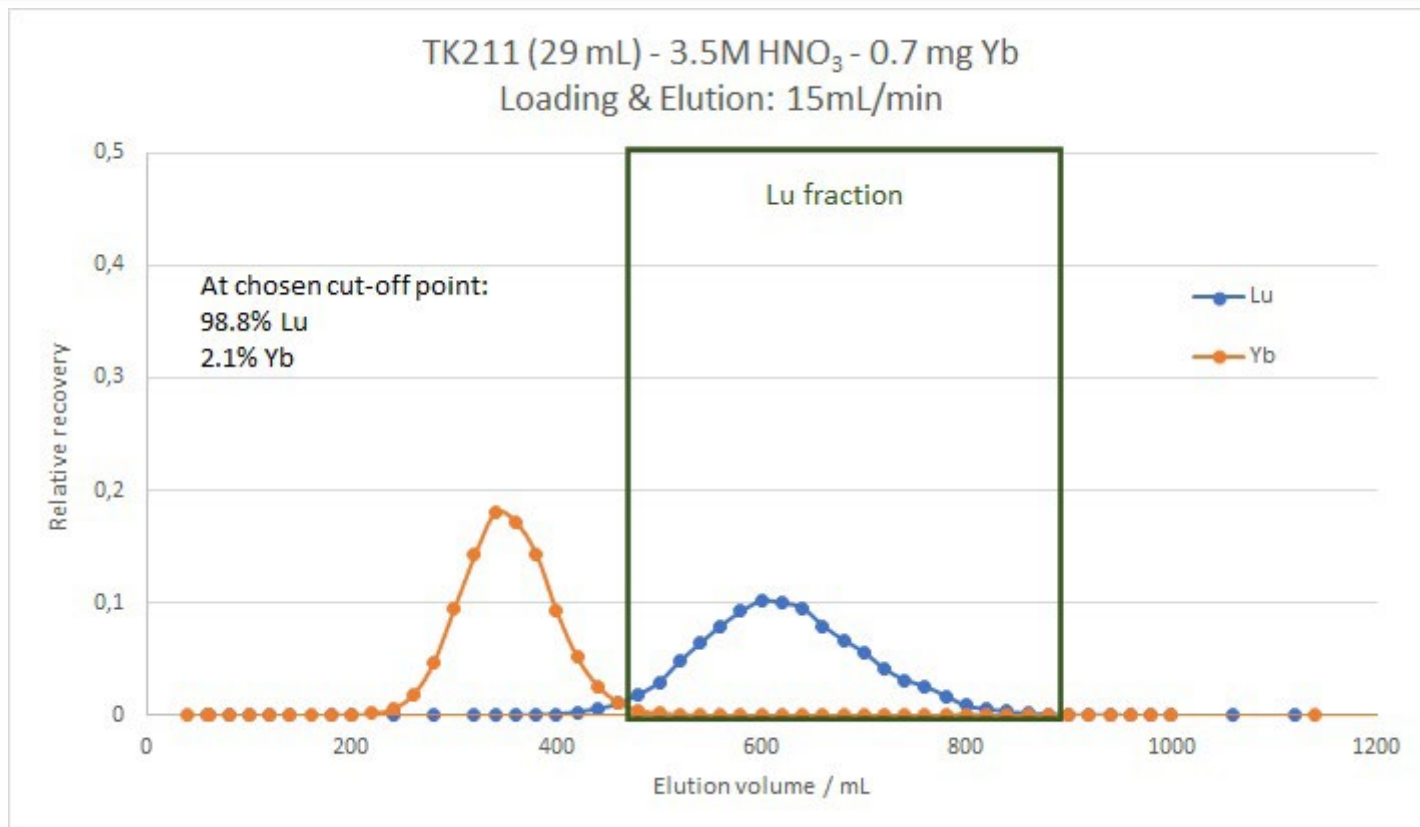
- Large tailing due to high Yb content => further optimisation on-going
- Improved separation through use of 1.25M HNO₃ / 10% EtOH (v/v)
- Higher Lu yield at similar residual Yb compared to LN2 based method
- Additional benefit from use of EtOH => improved radiolysis stability
- Online separation: switch at start of Lu fraction => ideally radiation detector driven

Lu separation from 500 mg Yb - TK212/TK221/TK212/TK211



- 2nd separation step on smaller TK212 (53 mL) after TK221 for conversion from high HNO₃ to dilute HCl on TK221
- Separation with e.g. 1.25M HNO₃ (with or without 10% EtOH)
- Direct loading of obtained Lu fraction onto TK211 Resin
 - Alternatively TK221/TK212 according to Horwitz et al.

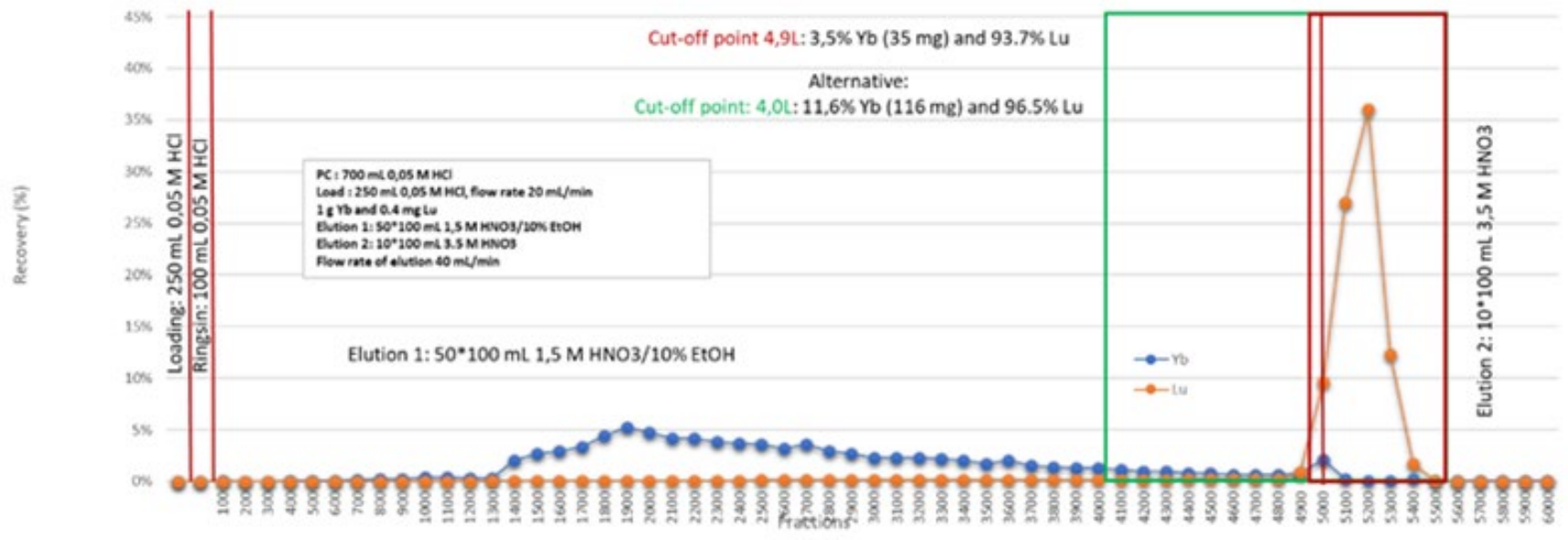
Lu separation from 500 mg Yb - TK212/TK221/TK212/TK211



- Lu / Yb separation on TK211 (29 mL) => Lu fraction directly loaded onto TK211 from TK212
- Overall Lu recovery of process approx 85%
- Low remaining Yb
- Flow rates may be optimized (slower flow will improve separation but will take time)
- Final step: concentration/conversion to $\leq 0.05\text{M}$ HCl on TK221, nitrate removal via A8

Ongoing work – further upscale

Separation of 1g Yb from Lu on 4 cm x 30 cm TK212 - 40 mL/min



- 375 mL TK212 column
- 1g Yb
- 40 mL/min flow rate
- After separation <120mg Yb left => TK221/second TK212