

# Cyclotron-based production of $^{68}\text{Ga}$ , $[^{68}\text{Ga}]\text{GaCl}_3$ , and $[^{68}\text{Ga}]\text{Ga-PSMA-11}$ from a liquid target -V 1.0

## 03/12/2024-TKI-GA01

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### 1. Scope

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This method aims to separate  $^{68}\text{Ga}$  from liquid Zn targets in order to obtain a pure isotope suitable for labelling.

### 2. Summary

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The reported method is based on the publication of Rodnick et al. (Rodnick et al., 2020). The main goal was to extract  $^{68}\text{Ga}$  from liquid Zn targets. Moreover, the main focus was not only to obtain an interference-free Ga product but also a product suitable for radiolabelling reactions (e.g. with PSMA-11). For the chemical separation of Ga from matrix elements, a cassette-based approach using two different cartridges sequentially, ZR Resin (Triskem International, 2016) and TK200 Resin (Triskem International, 2018), was employed. By using this approach, the on-line cartridge conditioning, the use of smaller amounts of acid and the exclusion of organic solvents and/or base-mediated pH adjustments was achieved. The complete radiochemical separation procedure led to chemical yields ranging from 87% to 93%. The final product was obtained in around 1-2 mL water plus dilute HCl to achieve desired formulation, suitable for further radiolabelling.

### 3. Significance of use

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This is a fast and automated method to obtain highly pure  $^{68}\text{Ga}$  from liquid Zn targets using a cassette-based approach. The same cartridge combination (ZR Resin and TK200 Resin) may also be used for solid Zn targets, however conditions need to be optimized accordingly.

### 4. Interferences

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Since Zn is the target element which will be irradiated to produce  $^{68}\text{Ga}$ , Zn was the main interference investigated. Additionally, enriched Zn liquid targets may contain some iron (1-3 ppm) which have also to be considered in regards of the chemical separation.

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### 5. Apparatus

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- Analytical balance -0.0001 g sensitivity
- Cassette-based synthesizer. In the publication this method is based on a FASTlab system is used, however other systems (AiO, Alceo,...) may also be employed
- Peristaltic pump
- Cassette reagent vials
- Cassette collection vials
- $\text{N}_2$  gas
- ICP-AES or other measurement equipment to quantify the different metal concentrations

### 6. Reagents

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#### a. Reagents

Unless otherwise indicated, all references to water should be understood to mean deionized distilled water. All reagents should be at least of analytical grade.

- 2 mL ZR column (50-100  $\mu\text{m}$ ) (TrisKem International, France)
- 2 mL TK200 column (50-100  $\mu\text{m}$ ) (TrisKem International, France)
- 69%  $\text{HNO}_3$
- 37%  $\text{HCl}$
- $\text{NaCl}$  salt
- Milli-Q water

#### b. Preparation of solutions

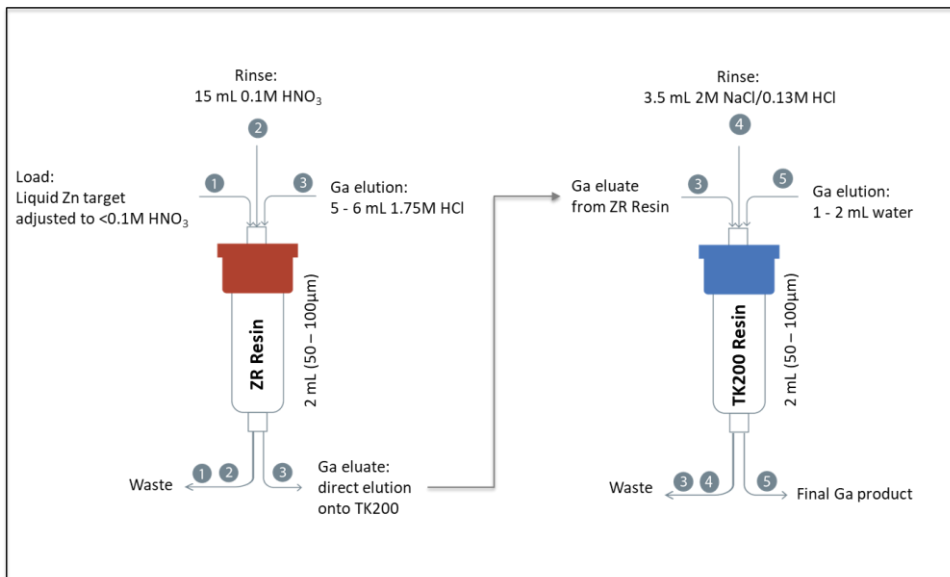
- **0.1 M  $\text{HNO}_3$ :** For 100 mL solution add around 70 mL deionized water at the bottom of the 100 mL flask and add slowly 625  $\mu\text{L}$  concentrated  $\text{HNO}_3$  (70%  $\text{HNO}_3$ ). Then, add water to the volumetric flask until the total volume. Mix thoroughly.
- **1.75 M  $\text{HCl}$ :** For 100 mL solution add around 50 mL deionized water at the bottom of the 100 mL flask and add slowly 14.58 mL concentrated  $\text{HCl}$  (37%  $\text{HCl}$ ). Then, add water to the volumetric flask until the total volume. Mix thoroughly.
- **2 M  $\text{NaCl}$ /0.13 M  $\text{HCl}$ :** For 100 mL solution add around 60 mL deionized water at the bottom of the 100 mL flask. Then, add 17.53 g  $\text{NaCl}$  and stir until  $\text{NaCl}$  has been completely dissolved. Finally, add water to the volumetric flask until the total volume. Mix thoroughly.

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### 7. Procedure

#### a. Graphical scheme



#### b. Sample preparation

1. Liquid Zn target irradiation on a cyclotron
2. Transfer the liquid Zn target solution into a 10 mL V-vial which is directly connected to the synthesizer
3. Dilution of the Zn target solution with water to  $<0.1\text{ M HNO}_3$
4. Load diluted Zn target onto the synthesizer cassette using nitrogen overpressure

#### c. Radiochemical separation

1. Conditioning of the two different cartridges:
  - a. Add 7 mL 0.1 M  $\text{HNO}_3$  to ZR Resin
  - b. Add 7 mL water and 4 mL 1.75 M  $\text{HCl}$  to TK200 Resin
2. Use nitrogen overpressure to load the target solution onto the cassette (ZR Resin cartridge)
3. Add 15 mL 0.1 M  $\text{HNO}_3$  onto ZR Resin in order to remove any possible interference
4. Add 5-6 mL 1.75 M  $\text{HCl}$  to elute Ga and directly load it onto the already preconditioned TK200 Resin
5. Add 3.5 mL 2 M  $\text{NaCl}$  in 0.13 M  $\text{HCl}$  onto the TK200 Resin
6. Finally, to elute Ga add 1-2 mL water followed by a few mL diluted  $\text{HCl}$  to yield desired formulation (e.g. 5 mL 0.1 M  $\text{HCl}$ )
7. Solution may be prepared e.g. for PSMA-11 labelling

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## 8. References

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- Triskem International. (2016). *Product sheet ZR Resin*. <https://www.triskem-international.com/catalog/products/resins-and-accessories/zr-resin/bl,product,424,0>
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